

Soft X-ray EPMA: choosing the right MACs

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● Use of soft X-rays for microanalysis

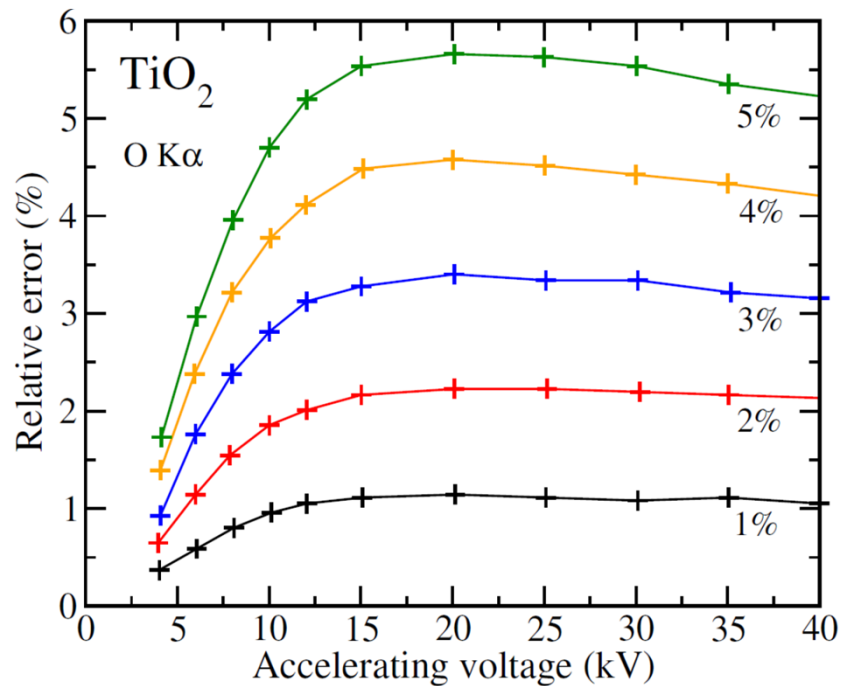
- Renewed interest due to advances in measurements of soft X-rays (synthetic multilayers, windowless EDS, SXES) and incorporation of FE guns in electron beam instruments
- K-lines of light elements (Li, Be, B, C, N, O, F). Play a critical role in many materials
- L-lines transition metals (Sc to Cu) and M-lines REEs: required for high-resolution EPMA (low voltage) of natural and technological materials

Difficulties

- Soft X-rays produced in electron transitions involving *valence electrons*, thus they are affected by *chemical bonding*
- Potential large errors in evaluated concentrations because matrix corrections do not include corrections for chemical bonding
- X-ray absorption: dominant correction

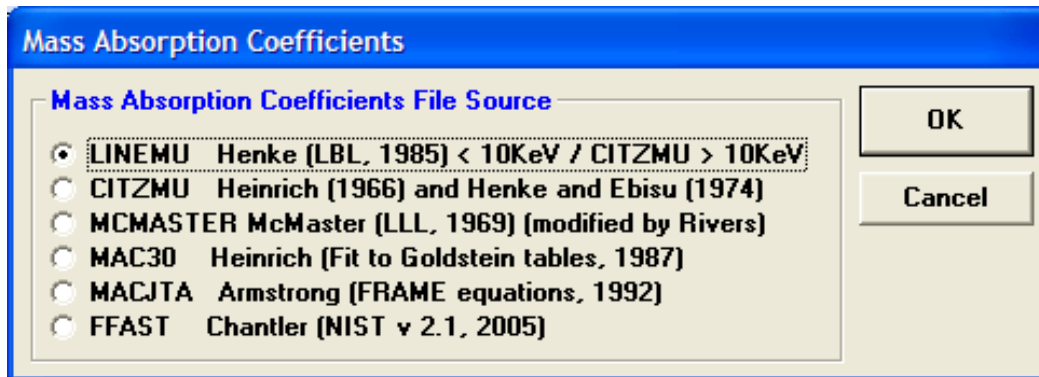
- **Key parameter: the mass attenuation coefficient (μ/ρ)**

- None of existing matrix corrections programs can provide accurate EPMA concentrations for soft X-rays without reliable MACs
- Concentrations sensitive to MAC choice for soft X-rays: a variation of 1% in the MAC leads to a variation of 1% in the evaluated concentration



Sensitivity of O concentration to MAC uncertainty for O K α X-rays emitted from TiO₂
MAC = 20,000 cm²/g

- Sources of MACs for soft X-rays



MAC choices in Probe for EPMA

- Semi-empirical databases: fits to experimental data + theoretical calculations to extrapolate across Z to fill the empty gaps
 - Henke et al. (1982), Henke et al. (1993), etc..
- Analytical formulas: fits to experimental data
 - Heinrich (1966), Heinrich (1986), Farthing-Walker (1990), etc..
- Theoretical (atomic) calculations:
 - NIST-FFAST Chantler (2005), Sabbatucci & Salvat (2016), etc..

Accuracy of MAC datasets for soft X-rays poorly documented: unclear if newer tabulations should replace older ones!

• Theoretical MAC calculations

For soft X-rays:

$$\frac{\mu}{\rho} \sim \frac{N_A}{A_M} \sigma_{\text{ph}}$$

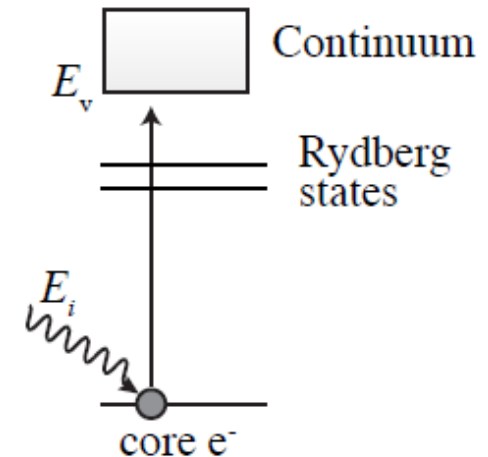
Photoelectric cross section

Photoelectric effect: a photon is absorbed by the target atom and an atomic electron is excited to an empty orbital, bound or free

$$d\sigma_{b;n_a\kappa_a m_a}^{\text{ion},1}(W, \xi) = \frac{(2\pi)^2 e^2}{W} \frac{k_b (\epsilon_b + m_e c^2)}{c\hbar} |M_{b;n_a\kappa_a m_a}^{\text{ion}}(W, \xi)|^2 d\hat{\mathbf{k}}_b,$$

$$M_{b;n_a\kappa_a m_a}^{\text{ion}}(W, \xi) = \xi \cdot \langle \psi_{k_b m_{Sb}}^{(-)} | \bar{\alpha} \exp(i\mathbf{k} \cdot \mathbf{r}) | \psi_{n_a \kappa_a m_a} \rangle$$

Sabbattucci & Salvat, Rad. Phys. Chem (2016) **121** 122



Additivity rule for compounds:

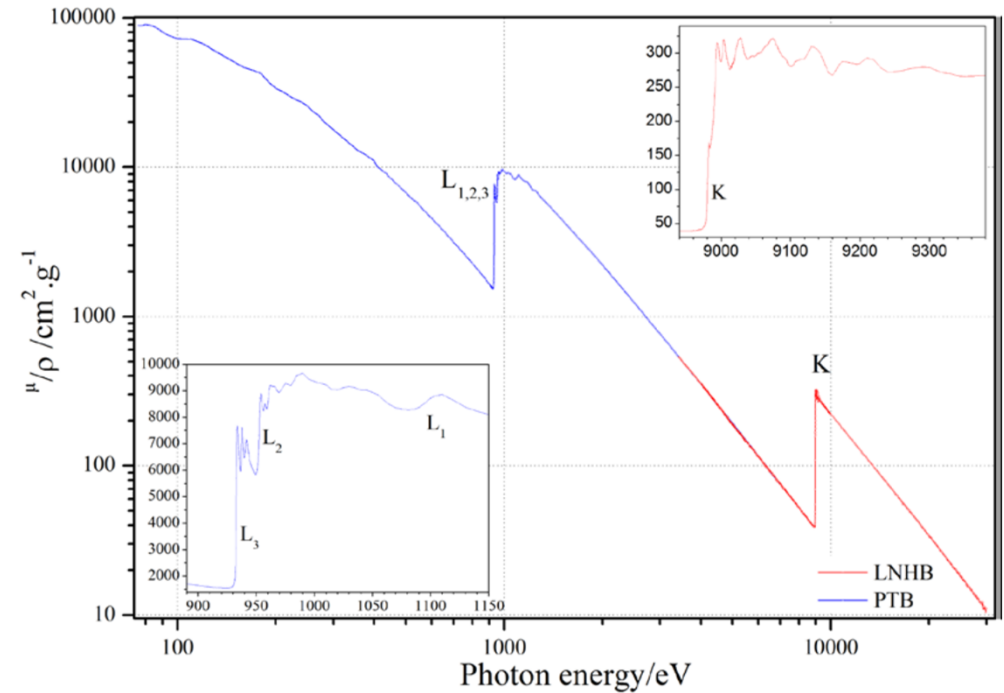
$$\left(\frac{\mu}{\rho}\right) = \sum_i c_i \left(\frac{\mu}{\rho}\right)_i$$

● Photoabsorption measurements

Experimentally determined by measuring the X-ray intensity I emitted through a layer of an absorbing element of mass thickness ρt :

$$\left(\frac{\mu}{\rho}\right) = -\frac{1}{\rho t} \ln \frac{I}{I_0}$$

- Saloman, Hubbell & Scofield (1988) compilation of experimental data published prior to 1998
- Ménesguen et al (2016, 2018a, 2018b): high-resolution photoabsorption measurements performed by an international initiative

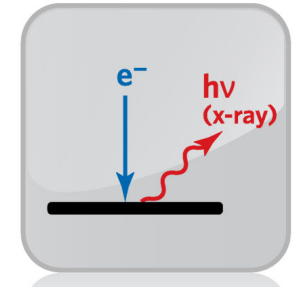


Ménesguen et al. (2016) Metrologia **53** 7

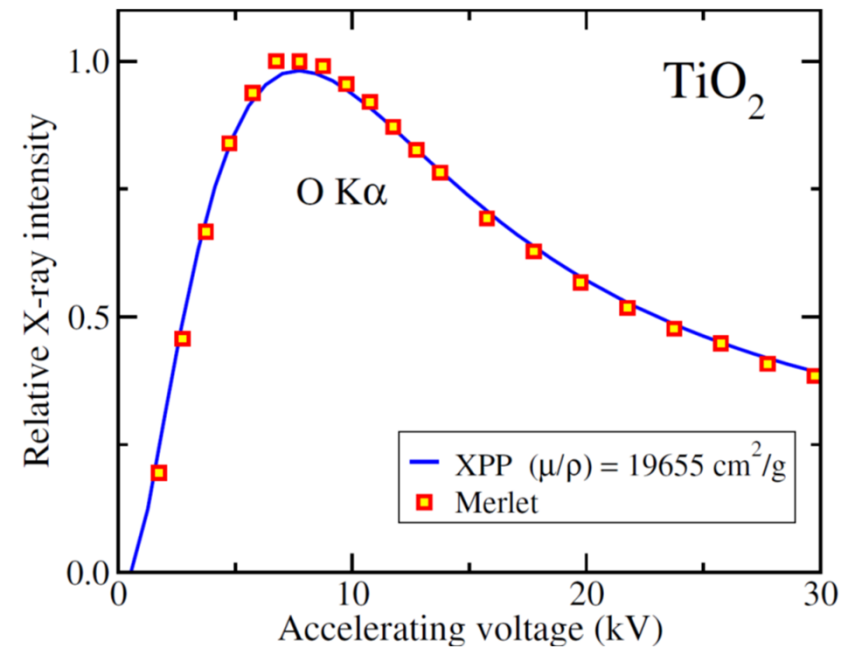
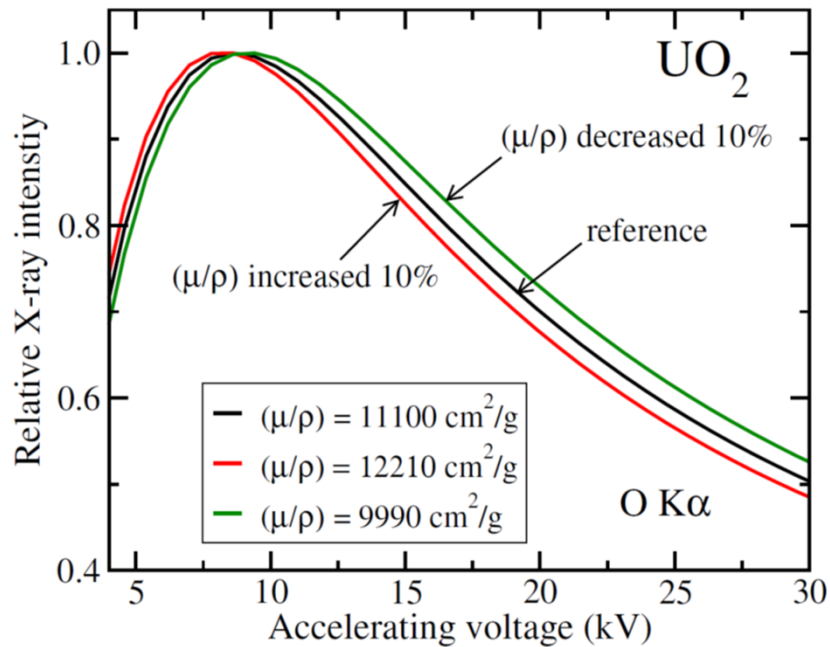
● Measuring the MAC with the EPMA

Assumption: MAC is the unknown in the quantification equation

$$I_i(E_0) \sim c_i \int_0^\infty \Phi_i(\rho z; E_0) \exp \left[-\frac{(\mu_i/\rho) \rho z}{\sin \chi} \right] d\rho z$$



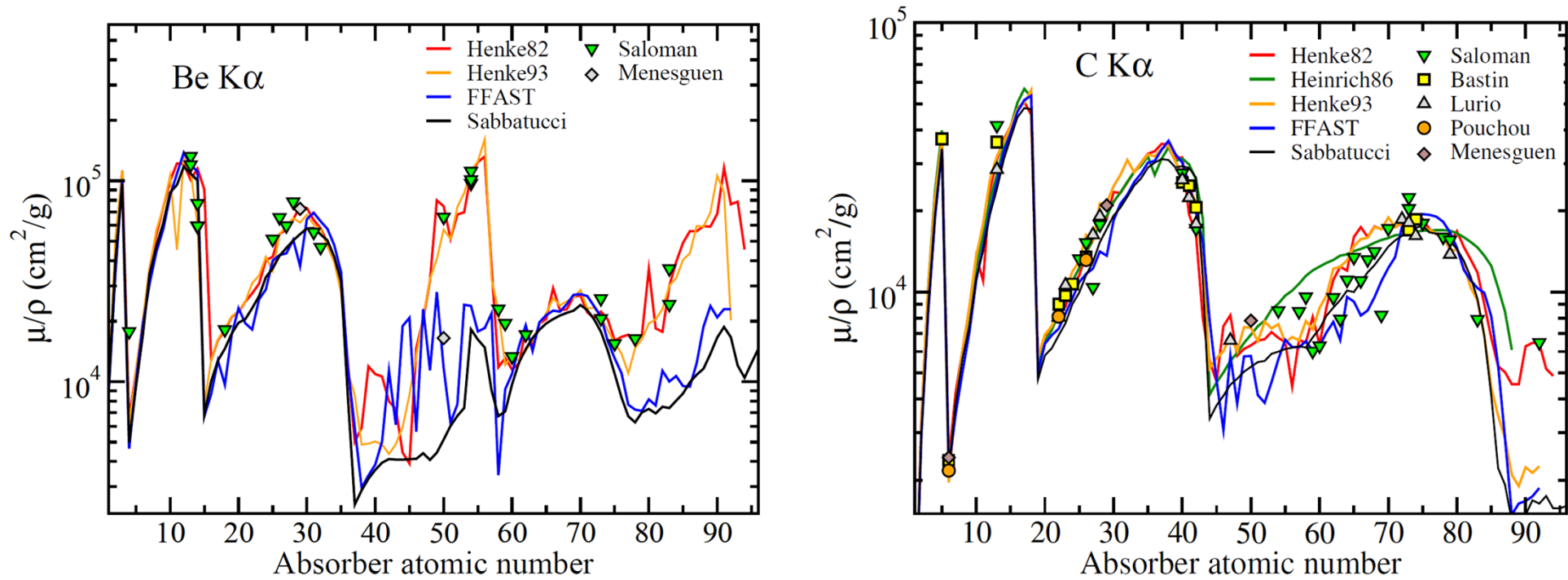
Software currently available: BadgerFilm and XMAC.
Technique developed by Pouchou (1998) and others



Sensitivity of the MAC to the X-ray intensity

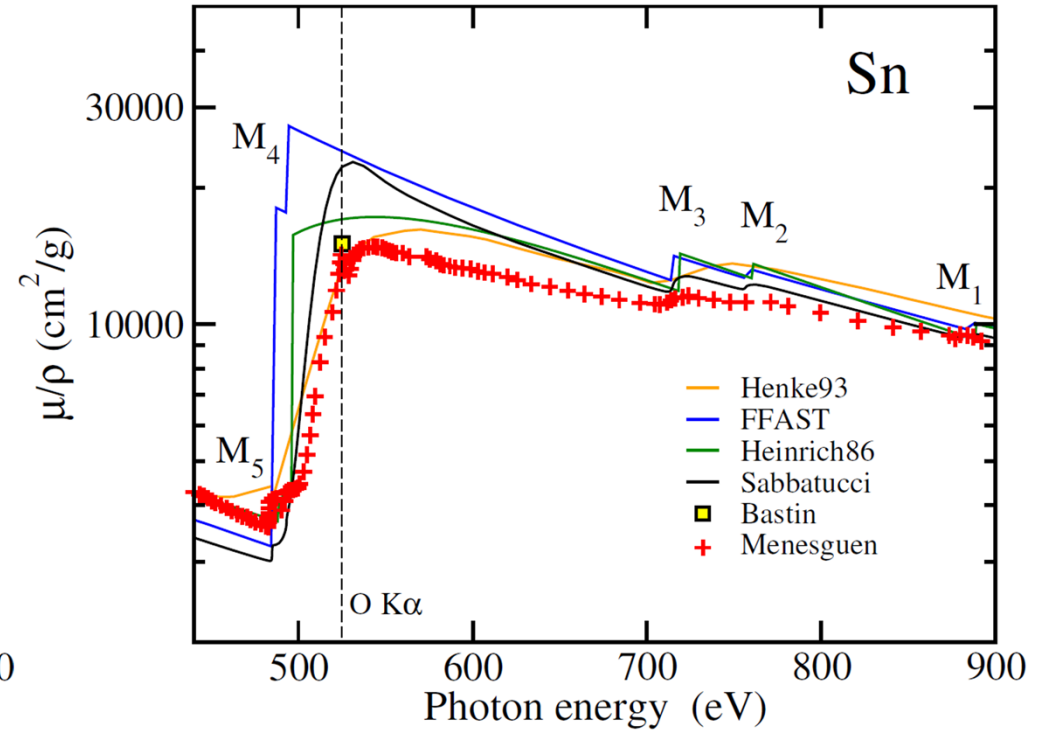
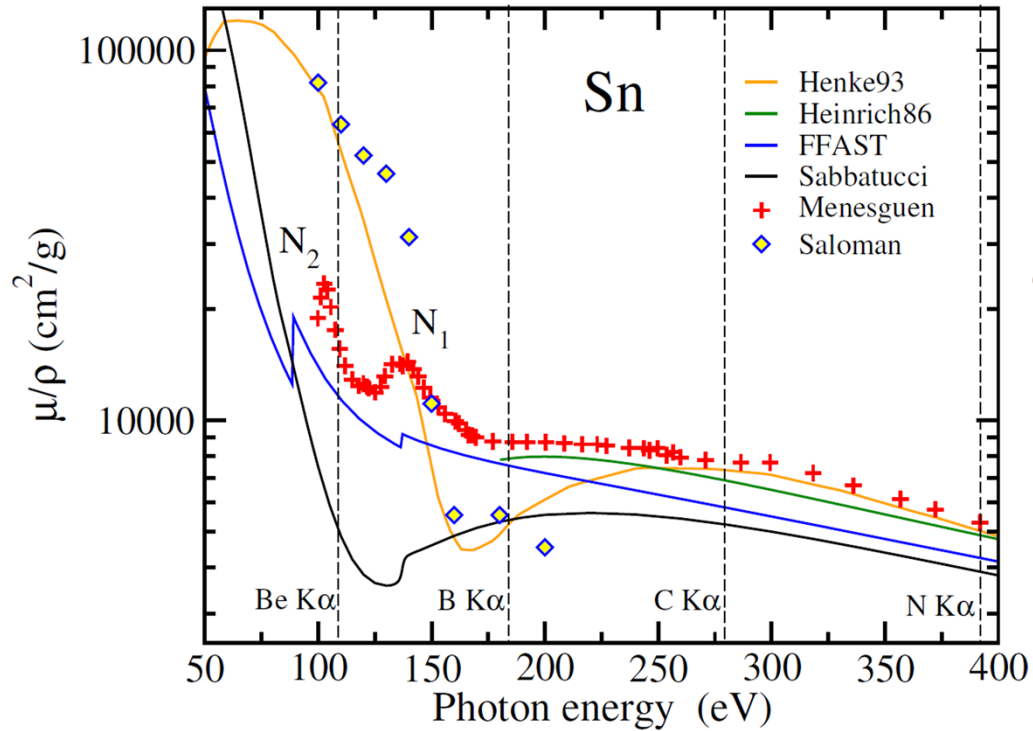
Experimental data: Kyser (1972), Bastin & Heijligers (1988, 1989, 1990, 1997), Pouchou & Pichoir (1988), Pöml & Llovet (2020), etc..

- Comparison with experimental data: Be and C emitters



- Significant discrepancies between the different tabulations
- Large spread of the experimental data (high-Z absorbers)

- Comparison with experimental data: Sn absorber



- EPMA measurements in good agreement with high-resolution photoabsorption data
- Discrepancies larger close to absorption edges and at lower X-ray energies

● Light elements: quantitative evaluation

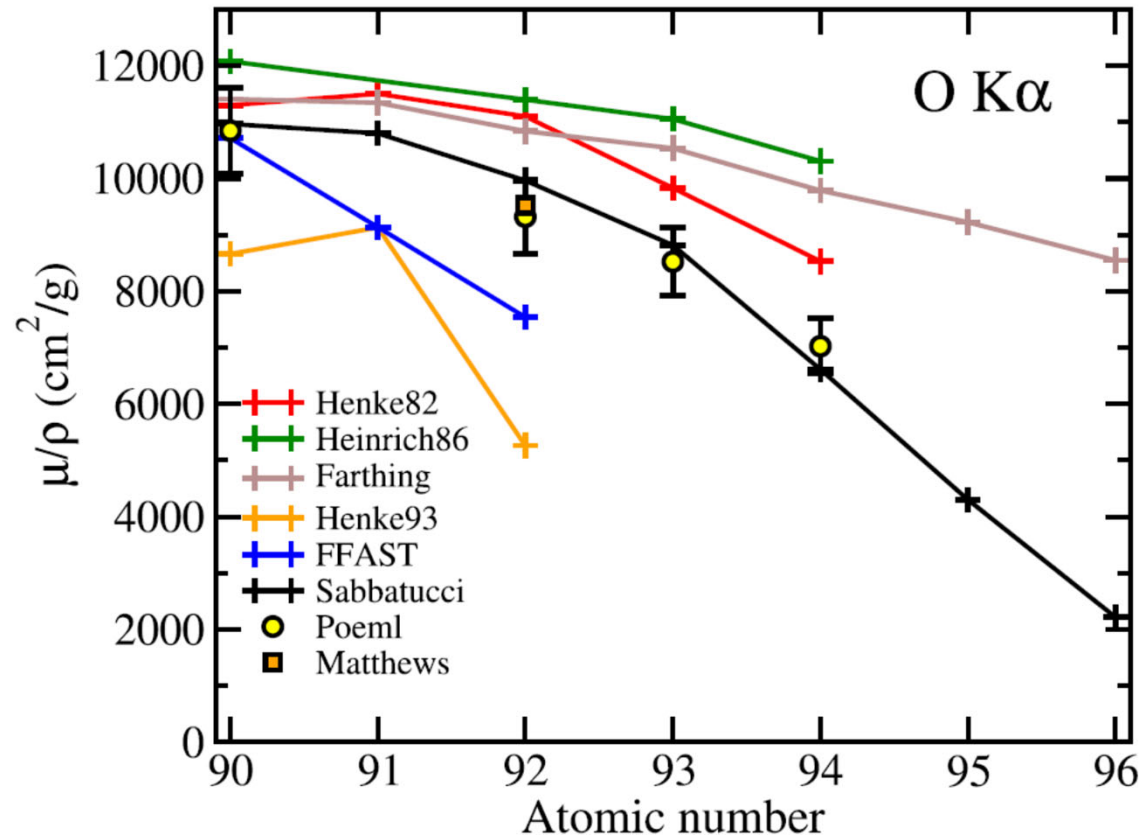
Root mean square
percentage deviation

$$RMS = \sqrt{\frac{1}{N} \sum_{i=1}^N (\Delta_i)^2} \quad \Delta_i = \frac{(\mu/\rho)_{\text{tab}} - (\mu/\rho)_{\text{exp}}}{(\mu/\rho)_{\text{exp}}} \times 100$$

Element	Henke82		Heinrich86		Henke93		FFAST		Sabbatucci	
	<i>RMS</i> (%)	<i>N</i>	<i>RMS</i> (%)	<i>N</i>	<i>RMS</i> (%)	<i>N</i>	<i>RMS</i> (%)	<i>N</i>	<i>RMS</i> (%)	<i>N</i>
Be	73.6	29	-	-	49.9	29	52.9	29	53.1	29
B	29.0	84	57.7	75	26.1	84	36.2	84	30.50	84
C	23.4	61	31.5	60	23.6	61	21.3	61	23.9	61
N	30.3	42	28.3	42	34.3	42	28.0	42	28.2	42
O	17.9	43	17.4	43	21.8	41	21.7	41	19.1	43
F	32.5	8	26.9	8	29.7	8	30.0	8	27.5	8
All	34.7	267	39.8	228	30.0	265	32.4	265	30.6	267

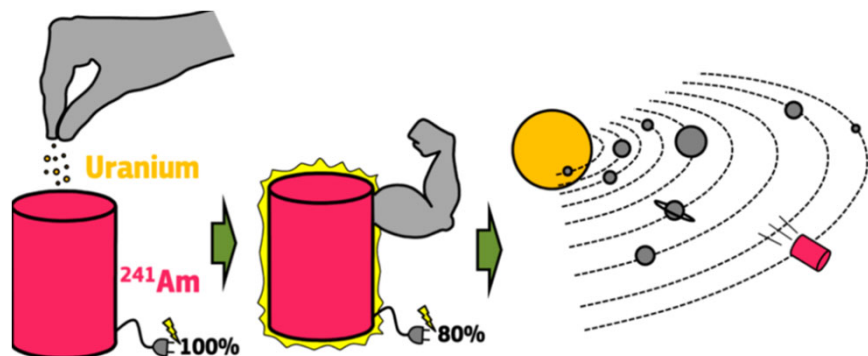
- Henke93 provides the lowest RMS, followed closely by Sabbatucci-Salvat
- More experimental data needed to draw definite conclusions, for high-Z absorbers

- Actinide absorbers



Recent EPMA based measurements for O K α X-rays in good agreement with Sabbatucci-Salvat's MACs

- EPMA analysis of U-doped Am oxide



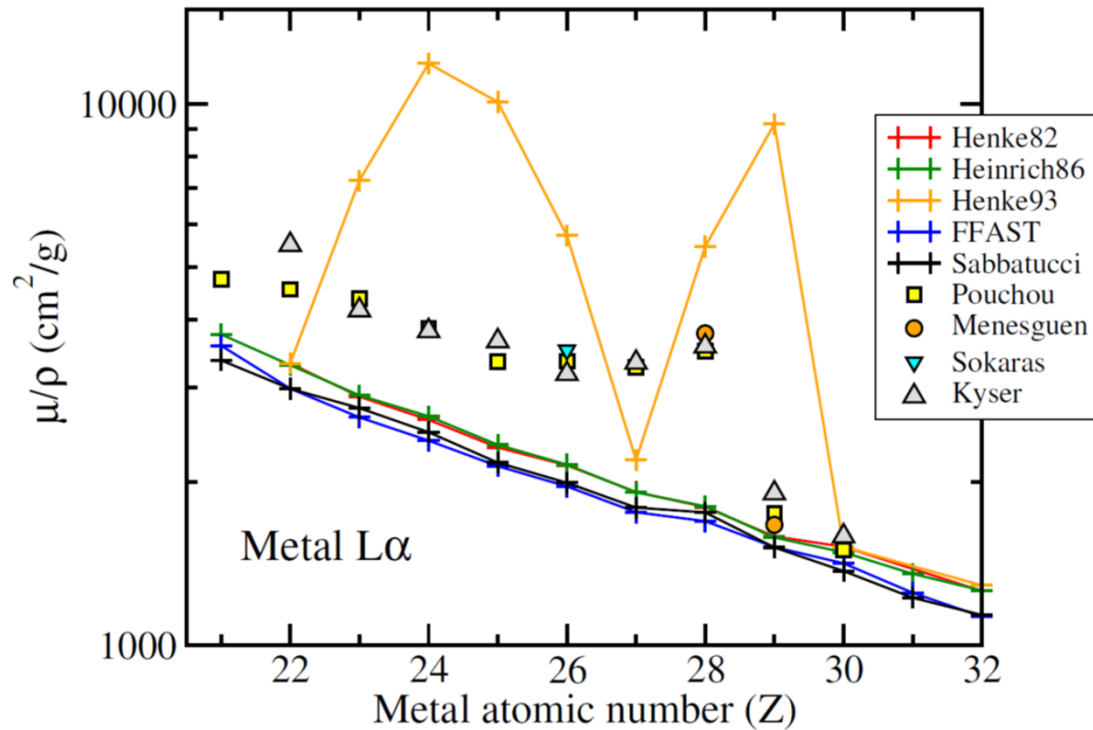
U-doped Am oxide developed as a radioisotope heater for space missions into deep space, where energy from the sun is not available

Vigier et al. *Inorg. Chem.* 2018, 57, 4317–4327

MAC source	U (wt.%)	Np (wt.%)	Pu (wt.%)	Am (wt.%)	O (wt.%)	Total
Farthing-Walker	11.2	5.01	1.57	72.2	17.1	107.2
Sabbatucci-Salvat	11.0	5.02	1.55	71.2	10.3	99.1

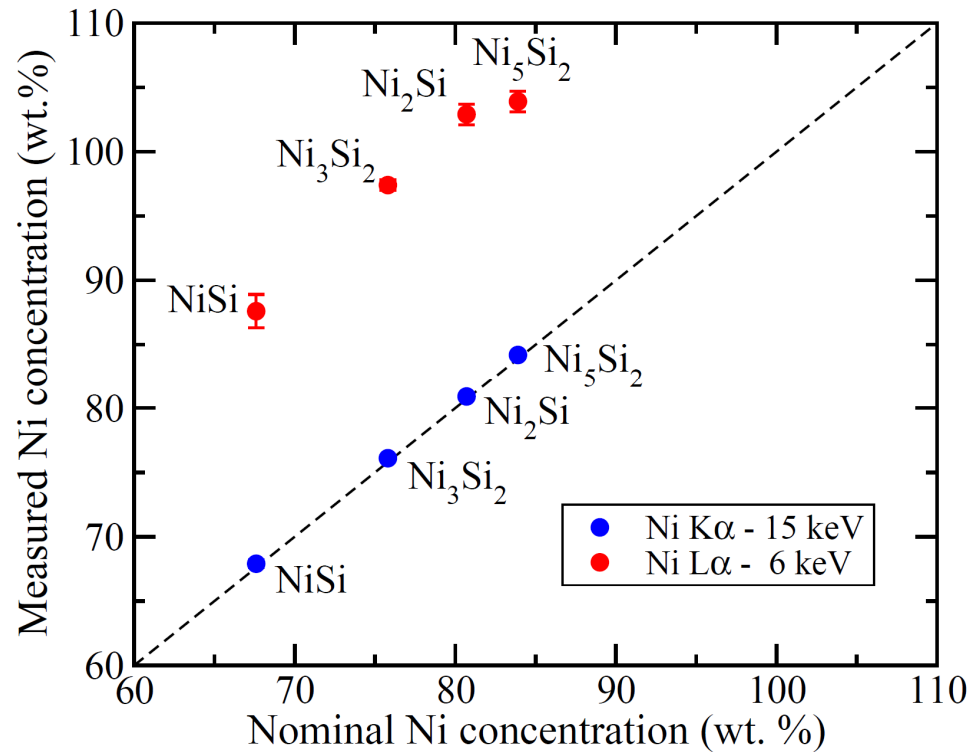
The use of the Sabbatucci-Salvat MACs provided a total closer to 100%

- Comparison with experimental data: transition metals



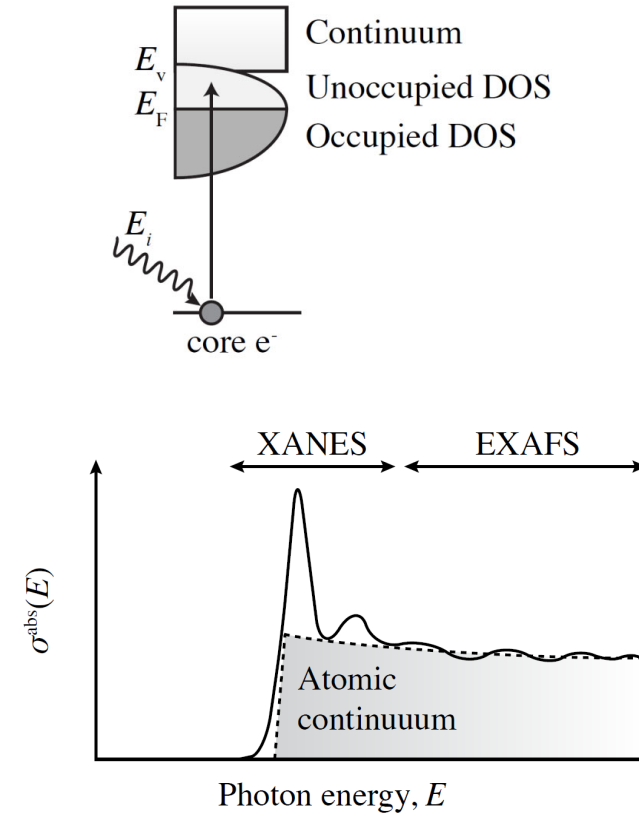
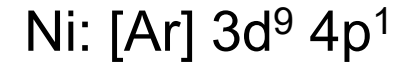
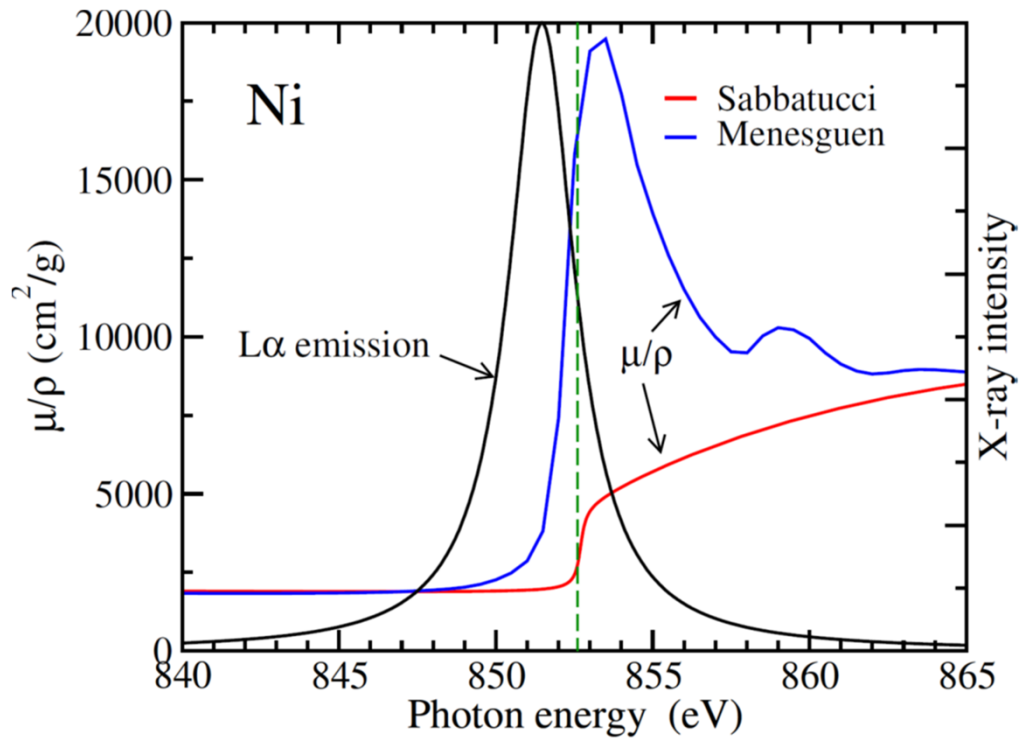
- Tabulations consistently underestimate experimental data by 30-40%
- Henke93 much higher than other tabulations (except for Ti, Co and Zn)

- EPMA analysis of Ni silicides using $L\alpha$ lines



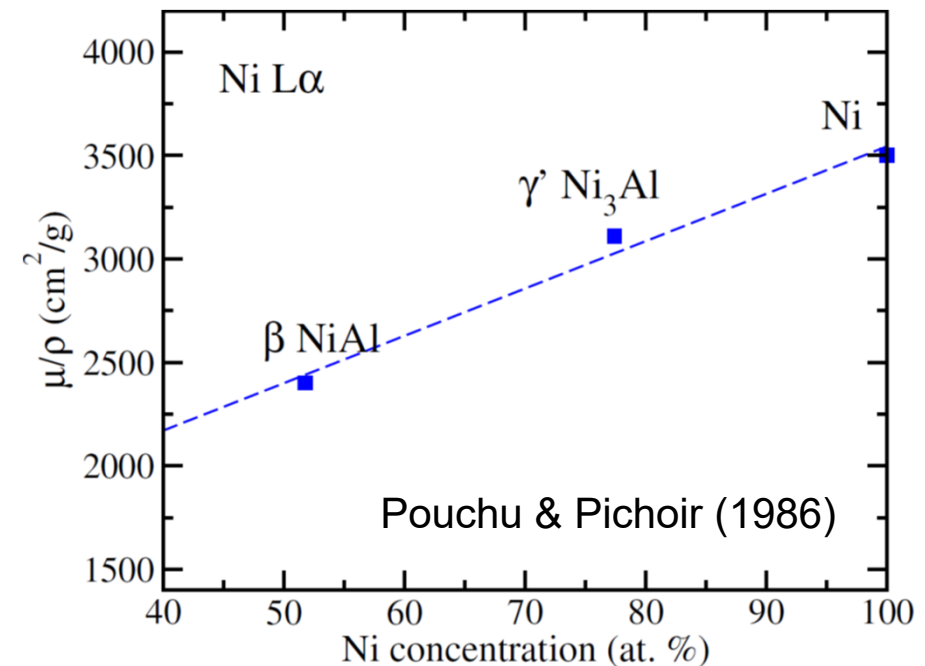
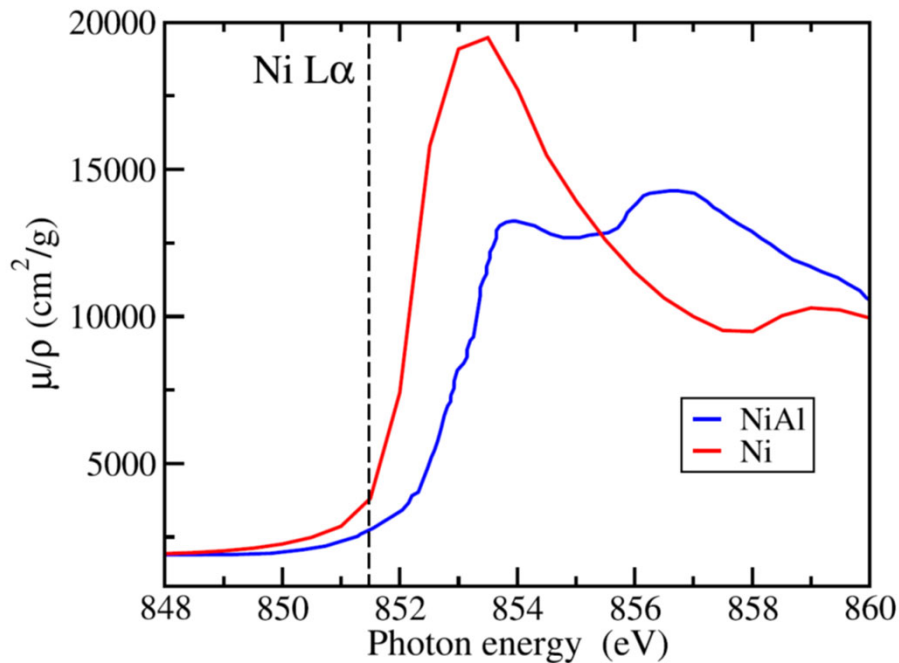
Ni $L\alpha$: deviation between measured and nominal concentrations of **20-30%**

- Absorption peaks at the edges



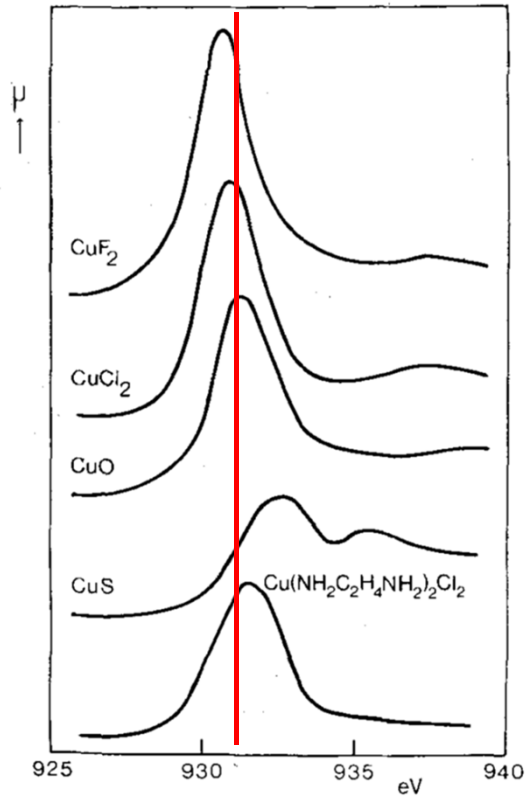
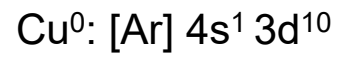
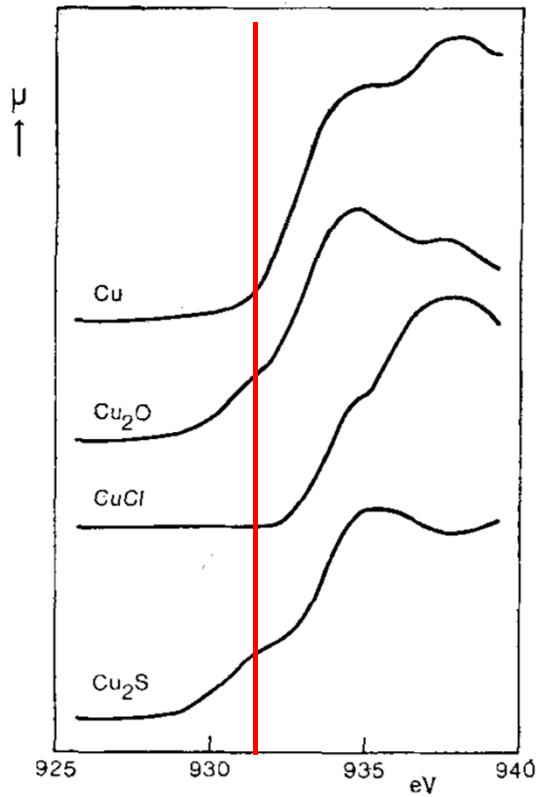
- Absorption peak: excitation of the unoccupied states of the 3d band, sensitive to the chemical state of the element
- Self-absorption: MAC increases rapidly across the X-ray line width (2.58 eV)

- **MAC sensitivity to chemical bonding: NiAl**



- Sensitivity to chemical bonding for NiAl: the filling of the 3d states by electrons from Al reduces the intensity of the absorption peak
- Breakdown of the additivity rule used to calculate MACs for compounds
- MAC measurements with the EPMA is a practical solution

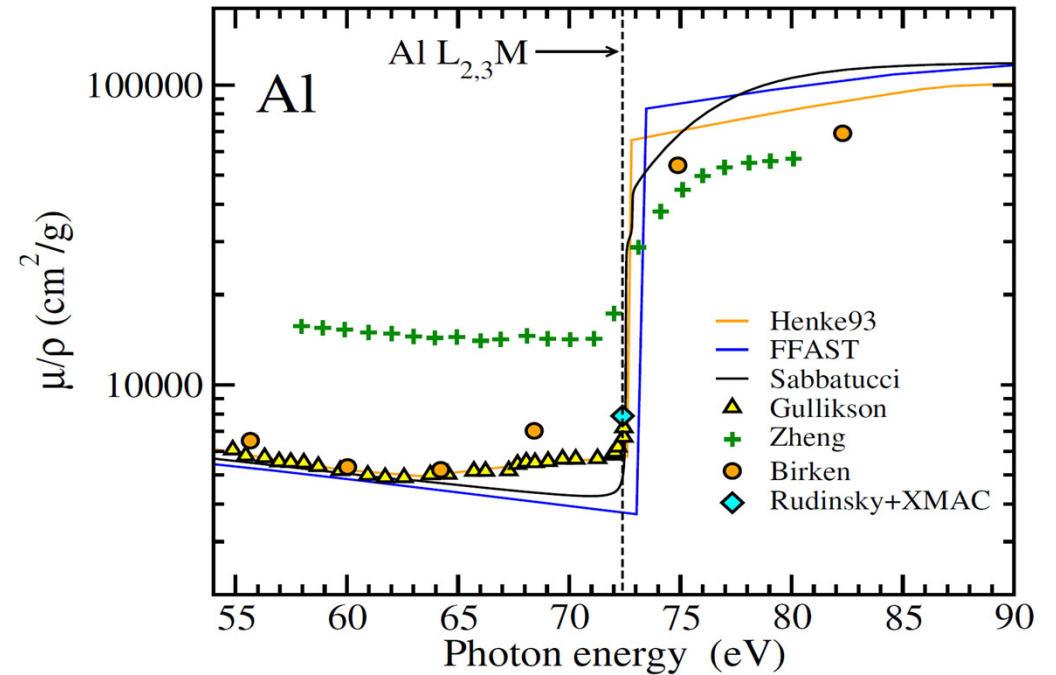
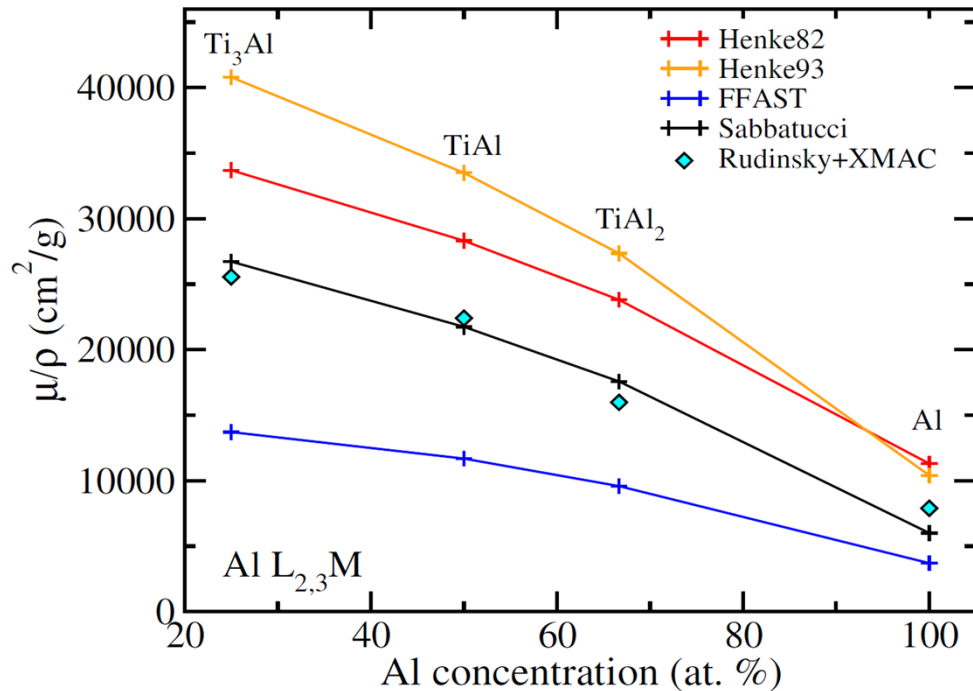
- **MAC sensitivity to chemical bonding: Cu compounds**



Kosler (1975)

Cu²⁺ compounds affected by self-absorption due to the presence of absorption peaks

● Ultra soft X-rays: Al L_{2,3}M X-rays (72.4 eV)



- MACs measured by Rudinsky et al (2020) using EPMA-SXES reprocessed with XMAC and Badgerfilm
- Good agreement with the MACs from Sabbatucci-Salvat for pure Al and several TiAl compounds

● Conclusions

- An assessment of the accuracy levels in MAC tabulations for soft X-rays has been presented by comparing tabulated MACs with experimental data available in the literature.
- The **1993 Henke** tabulation provides the lowest RMS value for light elements, but the theoretical MACs of **Sabattucci-Salvat** perform better if L-lines of transition metals and ultra-soft X-rays are considered
- EPMA allows determining the MAC when tabulated data is controversial or inexistent, especially near or between absorption edges.

Pöml P & Llovet X (2020) *Micros. Microanal.* **26** 194

Llovet X, Moy A & Fournelle JH (2022) *Micros. Microanal.* **28** 123

Llovet X, Pöml P, Moy A & Fournelle JH (2023) *Micros. Microanal.* **29** 540

