

Exploring Mars at the Nanoscale: Applications of transmission electron microscopy and atom probe tomography in planetary exploration

Luke Daly



Martin Lee

Àine O'Brien

Sammy Griffin

Benjamin Cohen

Lydia Hallis



University of Glasgow | School of Geographical & Earth Sciences



John Halpin

Will Smith

Sam McFadzean



University of Glasgow | School of Physics & Astronomy



Science & Technology
Facilities Council

Paul Bagot



UNIVERSITY OF
OXFORD



OxfordMaterials

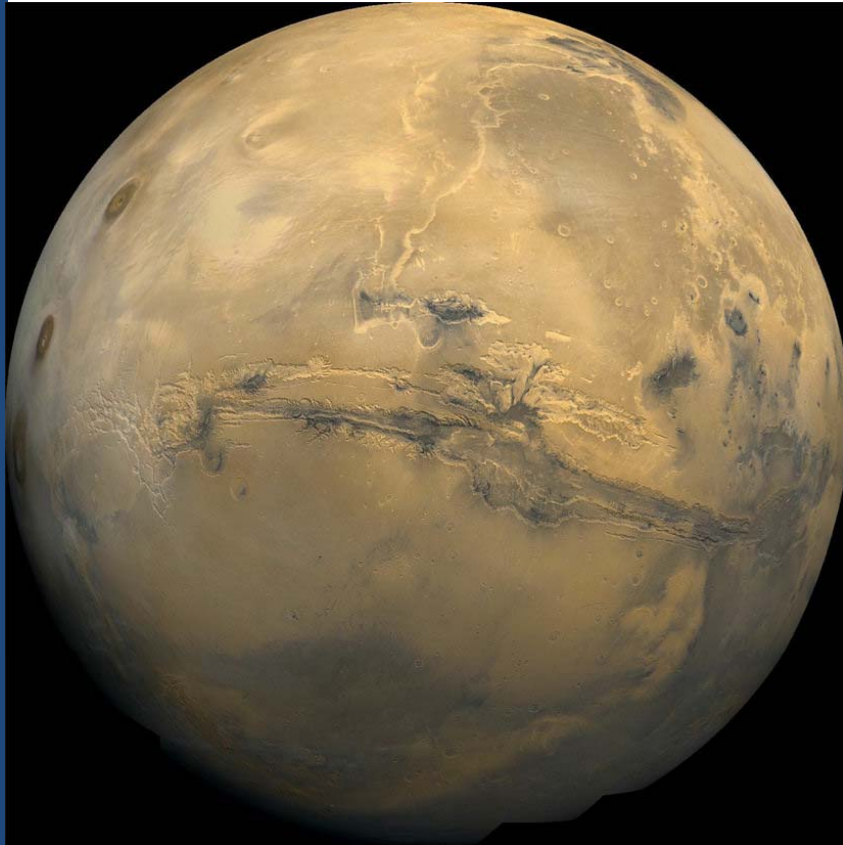
Julie Cairney

Ingrid McCarrol



THE UNIVERSITY OF
SYDNEY





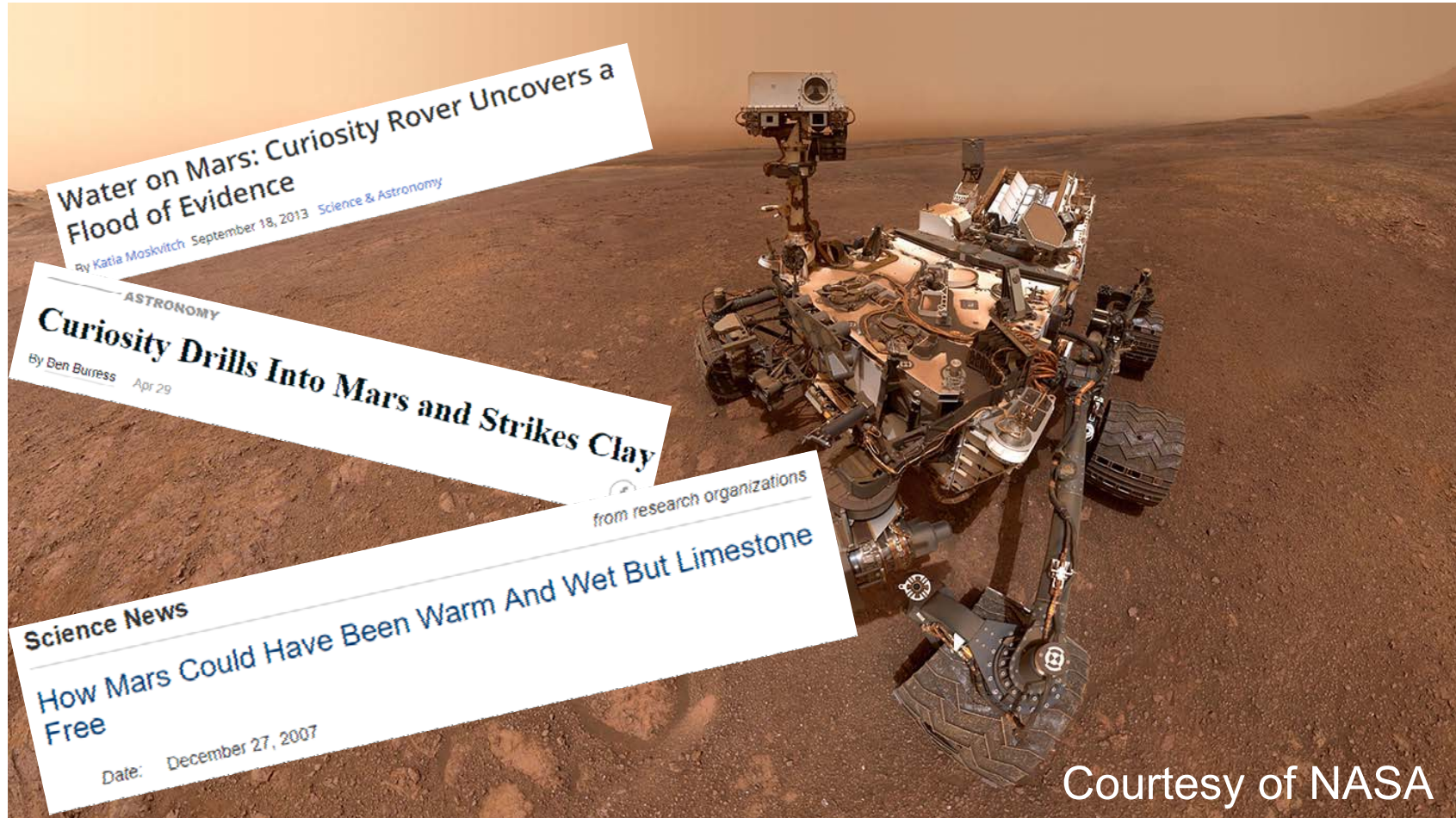
Courtesy of NASA

- Remote sensing indicates large fluvial systems were once active on the Martian Surface.
- Modern Mars is too cold for liquid water to be present at the surface.
- Could liquid water persist somewhere on Mars
- Martian meteorites provide a physical sample of the Mars' crust.



University of Glasgow

Evidence of Water on Mars : Curiosity



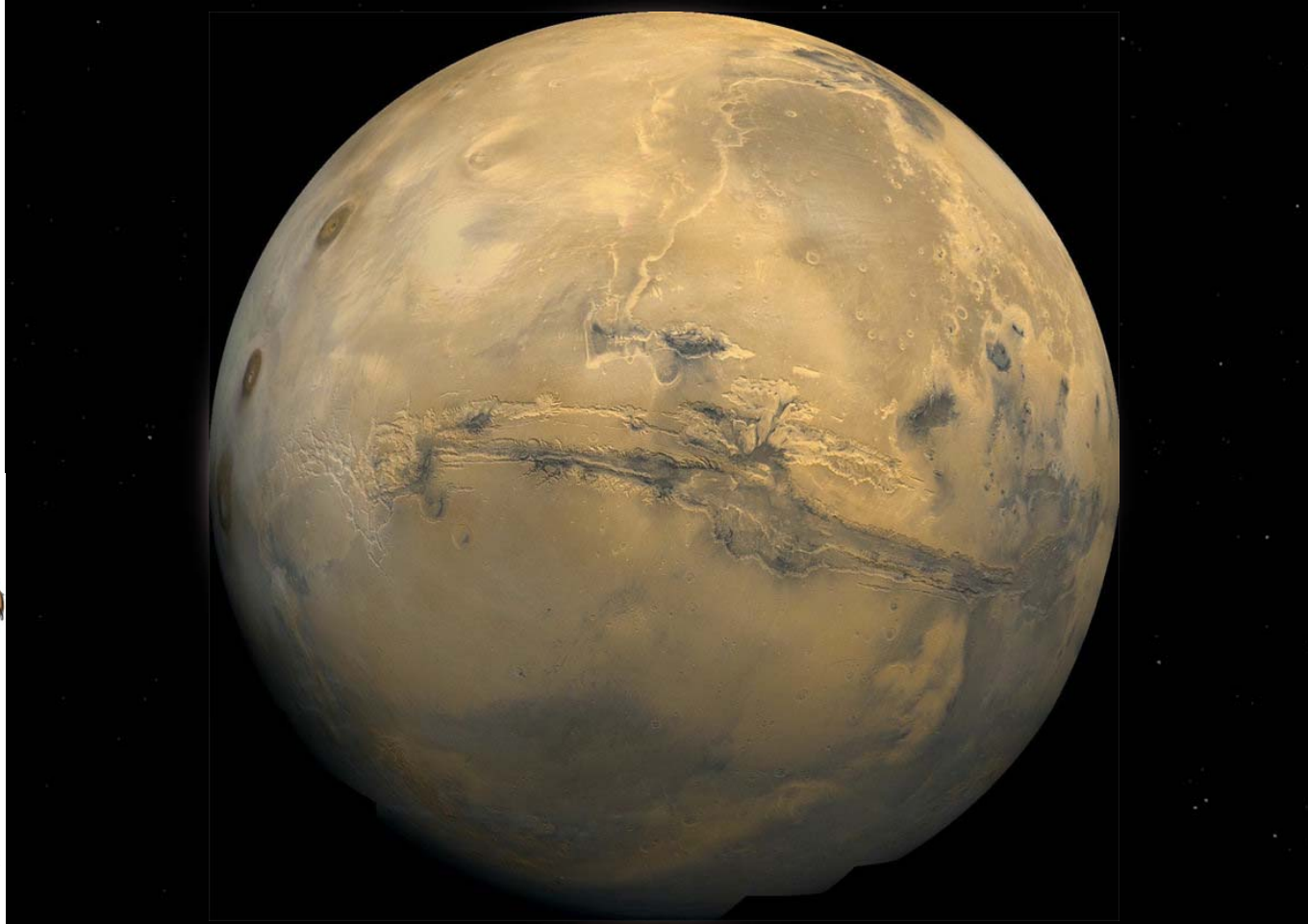


Courtesy of NASA

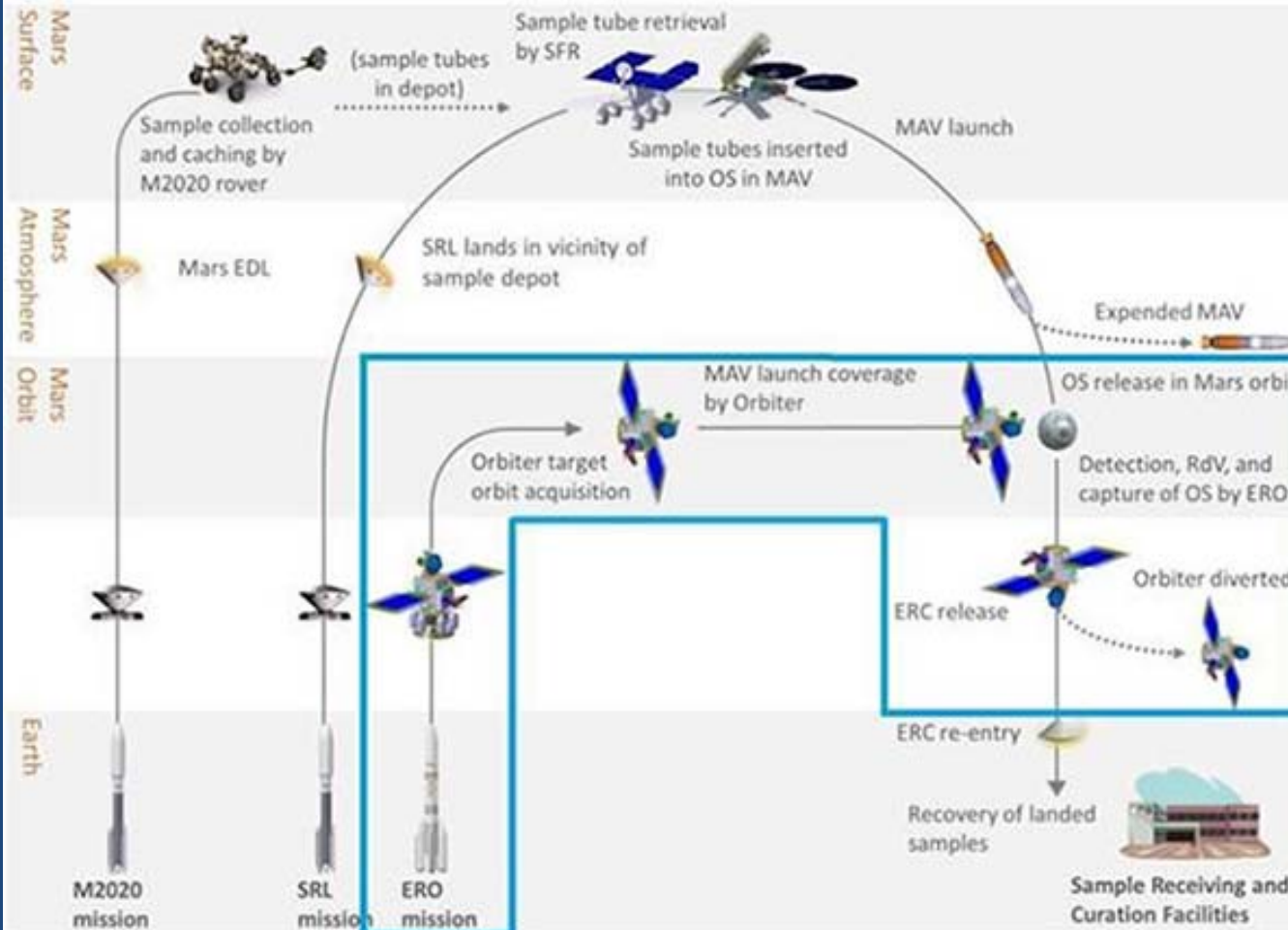


University
of Glasgow

Martian Oceans?



Courtesy
of NASA

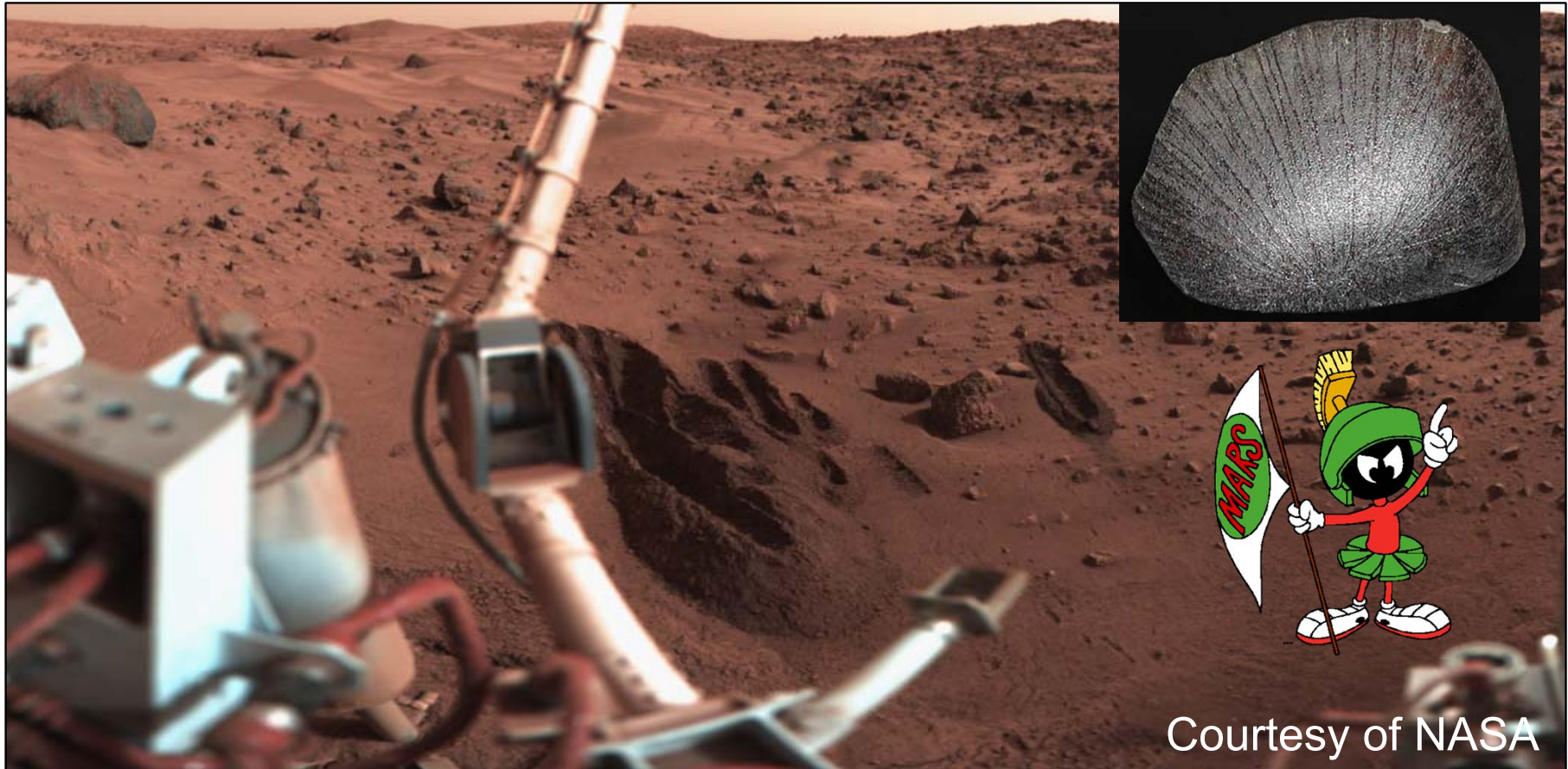


- Several missions in one
- Return ~1kg of Mars
- Must maximise research output from small sample volumes



University
of Glasgow

Viking 1: Landed on Mars, 20 July 1976



Courtesy of NASA

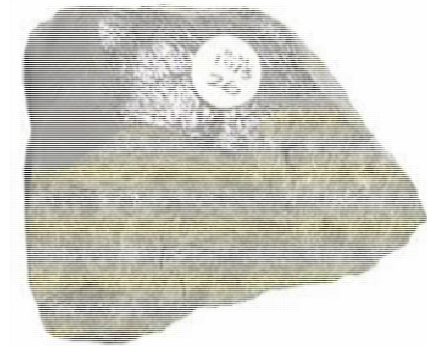
Lafayette

0.4 wt. % water



Nakhla

0.1 wt.% water



Nakhla

31°19' N., 30°21' E.

Abu Hommos, Alexandria, Egypt

Fell 1911, June 28, 0900 hrs

Synonym(s): *Abdel Malek, El Nakhla el Baharia*

Stone. Achondrite, Ca-rich. Nakhlite (ACANOM).

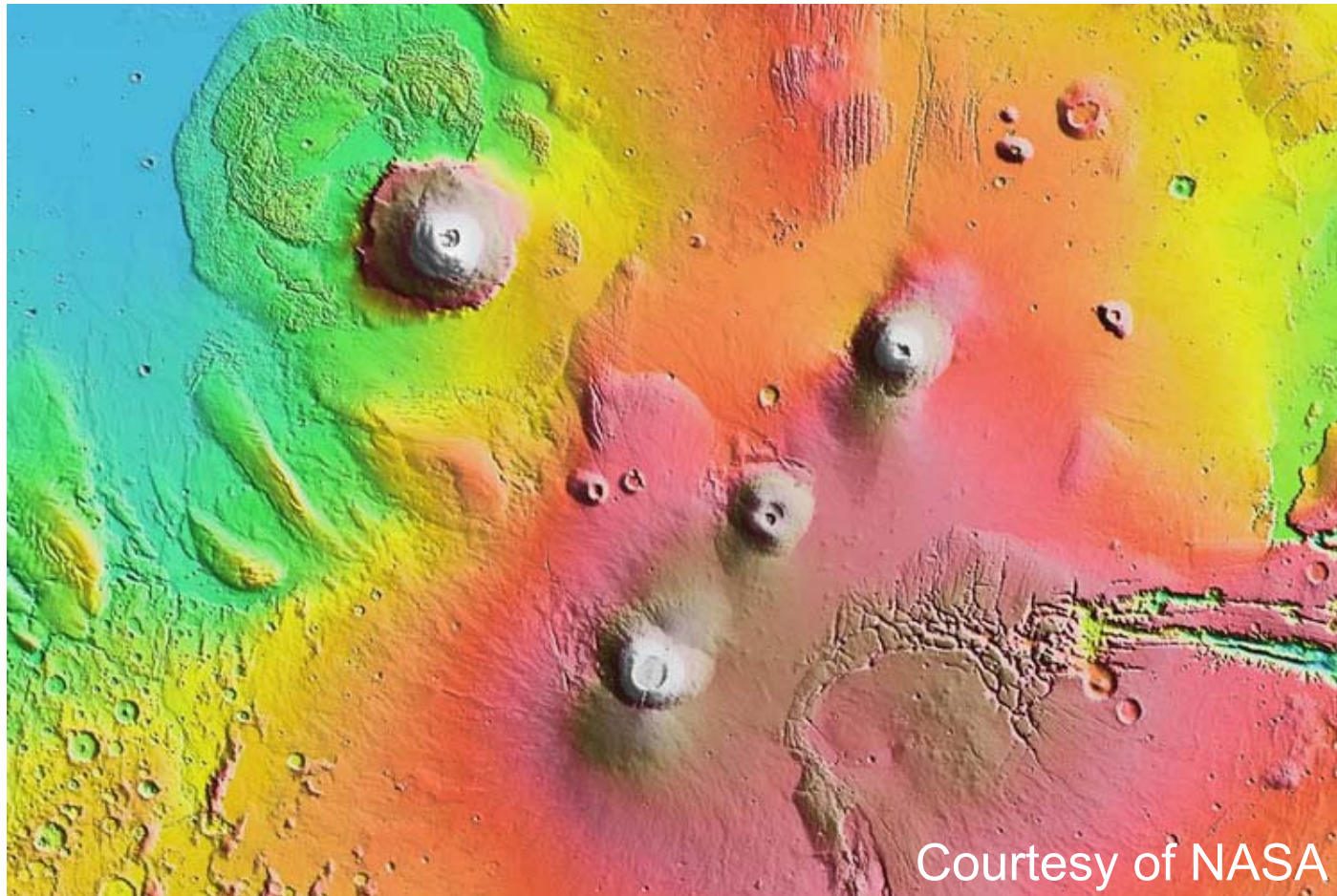
About 40 stones, of total weight about 40kg, and varying in weight from 1813g to 20g fell, after appearance of cloud and detonations, W.F. Hume, Cairo Scientific Journal, 1911, 5 (59), p.212. Mentioned, J. Ball, Geol. Reports, Surv. Dept. (Egypt), Cairo, 1912 (25). One of the stones killed a dog. Description and analysis, G.T. Prior, Min. Mag., 1912, 16, p.274. Consists mainly of a green diopside with some highly ferriferous olivine and a little feldspar. New analysis, 16.16 % total iron, T.S. McCarthy et al., Meteoritics, 1974, 9, p.215. Ar-Ar age, 1300, m.y., F.A. Podosek and J.C. Huneke, Geochimica et Cosmochimica Acta, 1973, 37, p.667. Rb-Sr

Nakhla

0.1 wt.% water



Nakhlite Meteorites history: Formation



(rim and central uplift)



Not to scale

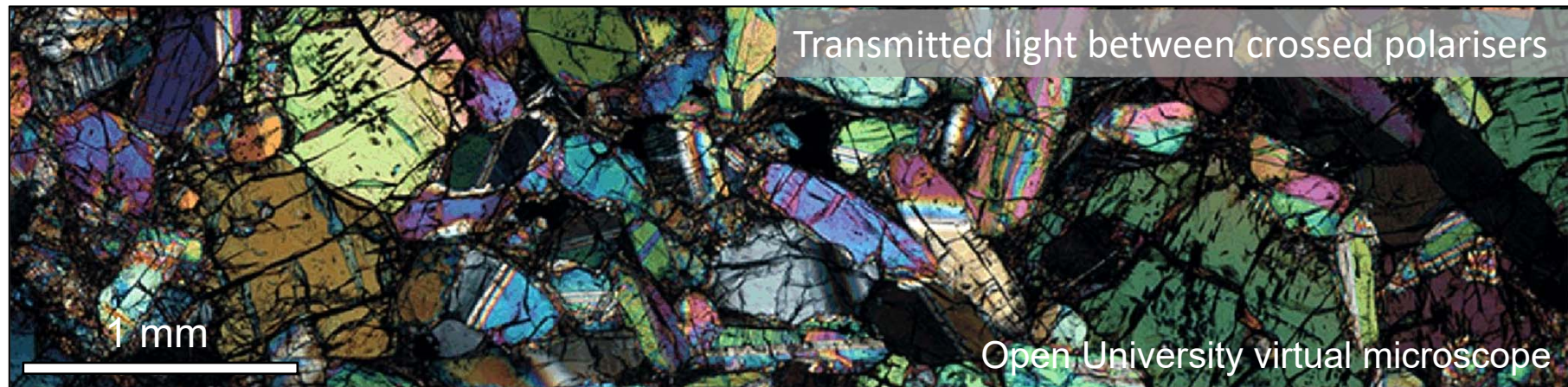
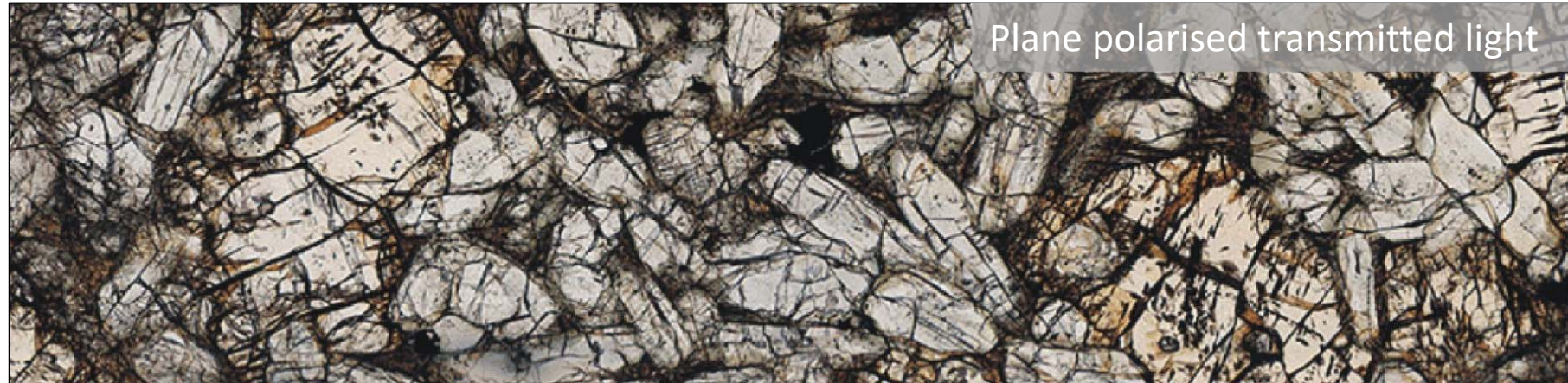
Courtesy of NASA

Nakhlite Meteorites history: Formation



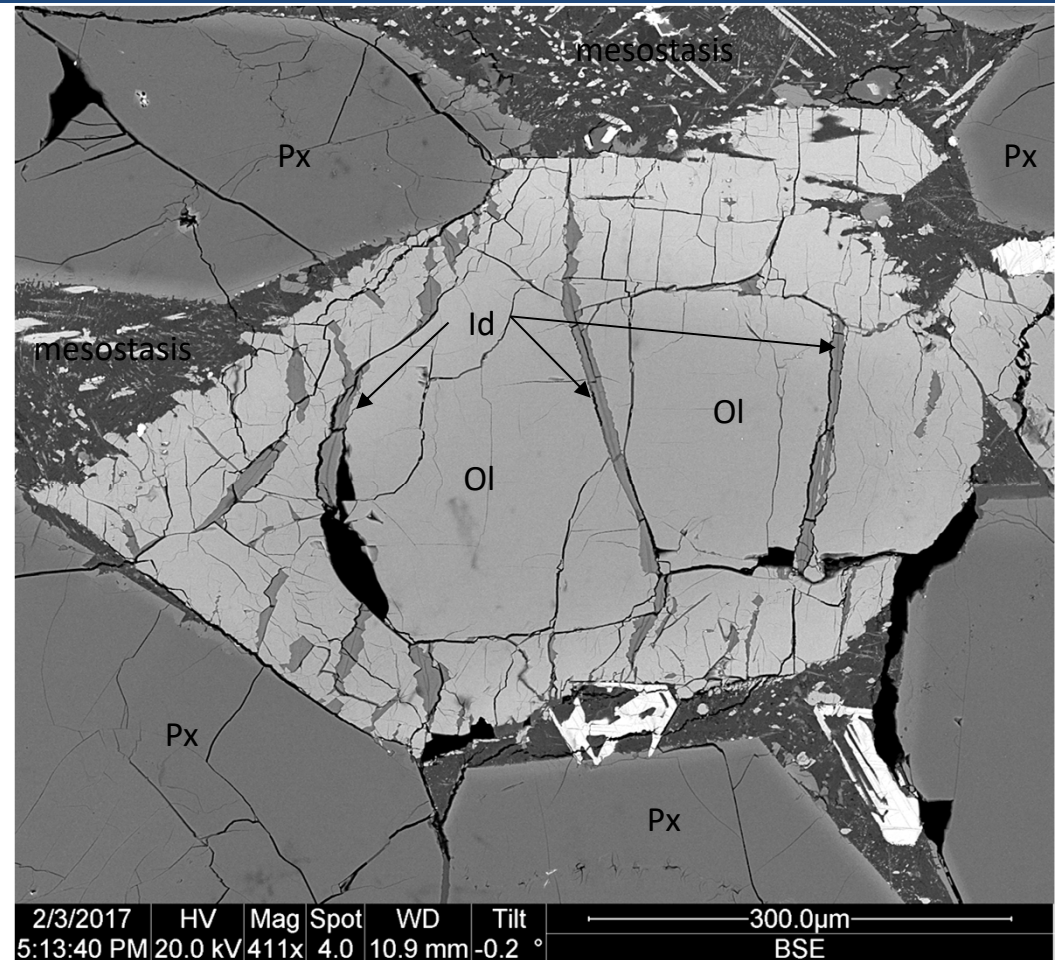
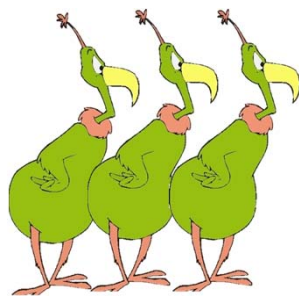
- 1400-1300 million year old volcanic rocks.
- Made mainly of the Mg-Fe-(Ca) silicate minerals augite and olivine.

Nakhlite meteorite in thin section



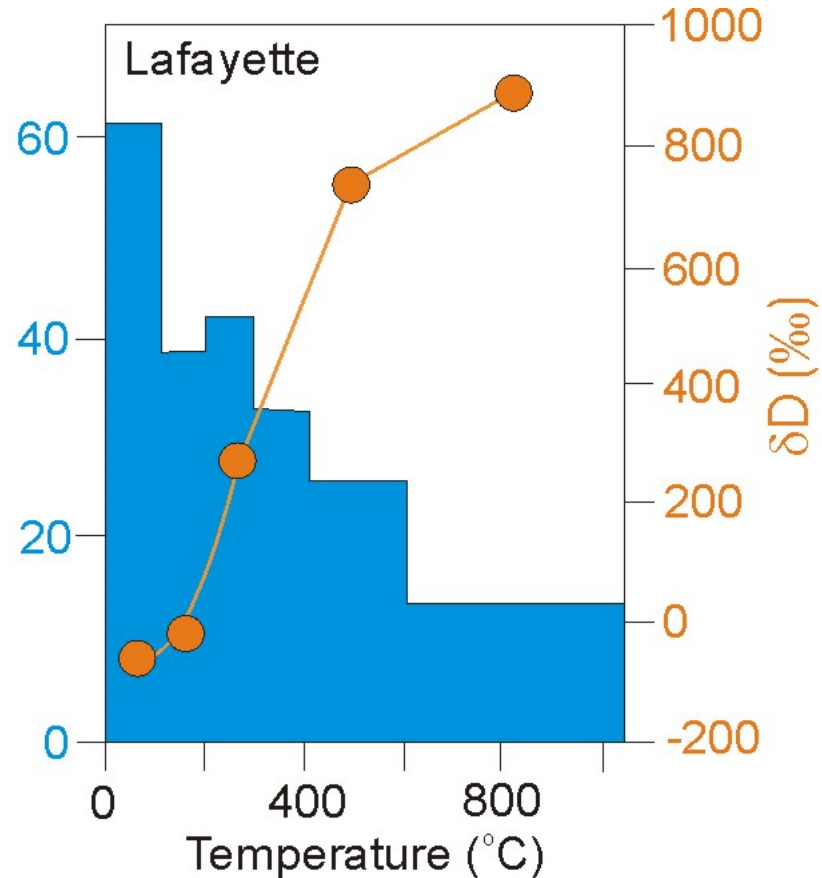
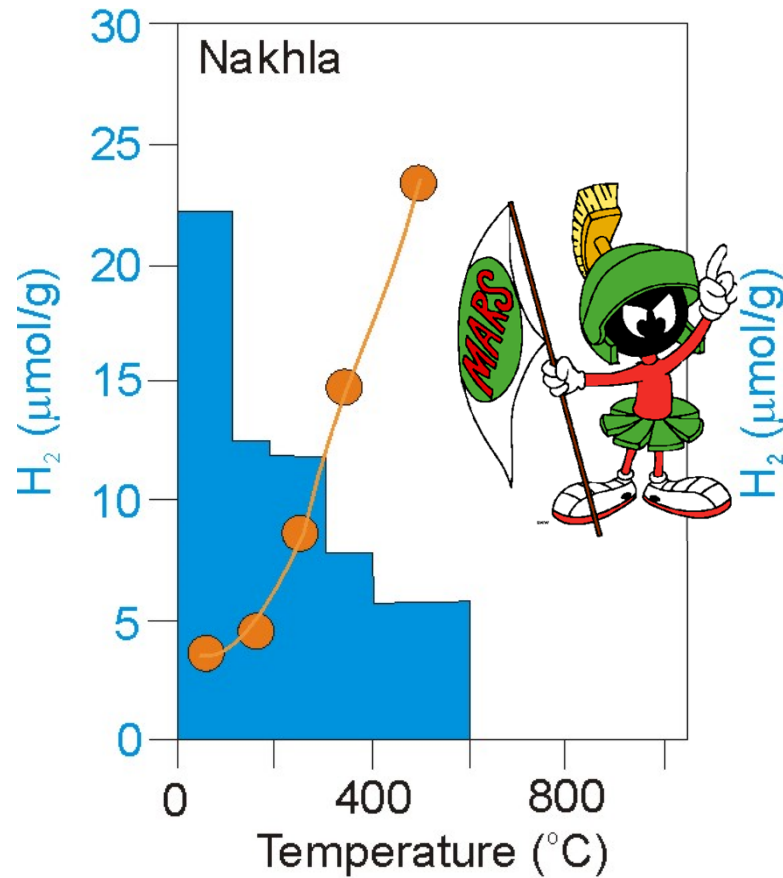
Nakhlite meteorites history: evidence of water

- Nakhlite meteorites from Mars contain precious samples of the planet's water;
- Evidence for recent ~633 Ma aqueous activity
- The heat source for these fluids is unknown





Demonstrating a Martian provenance for the water

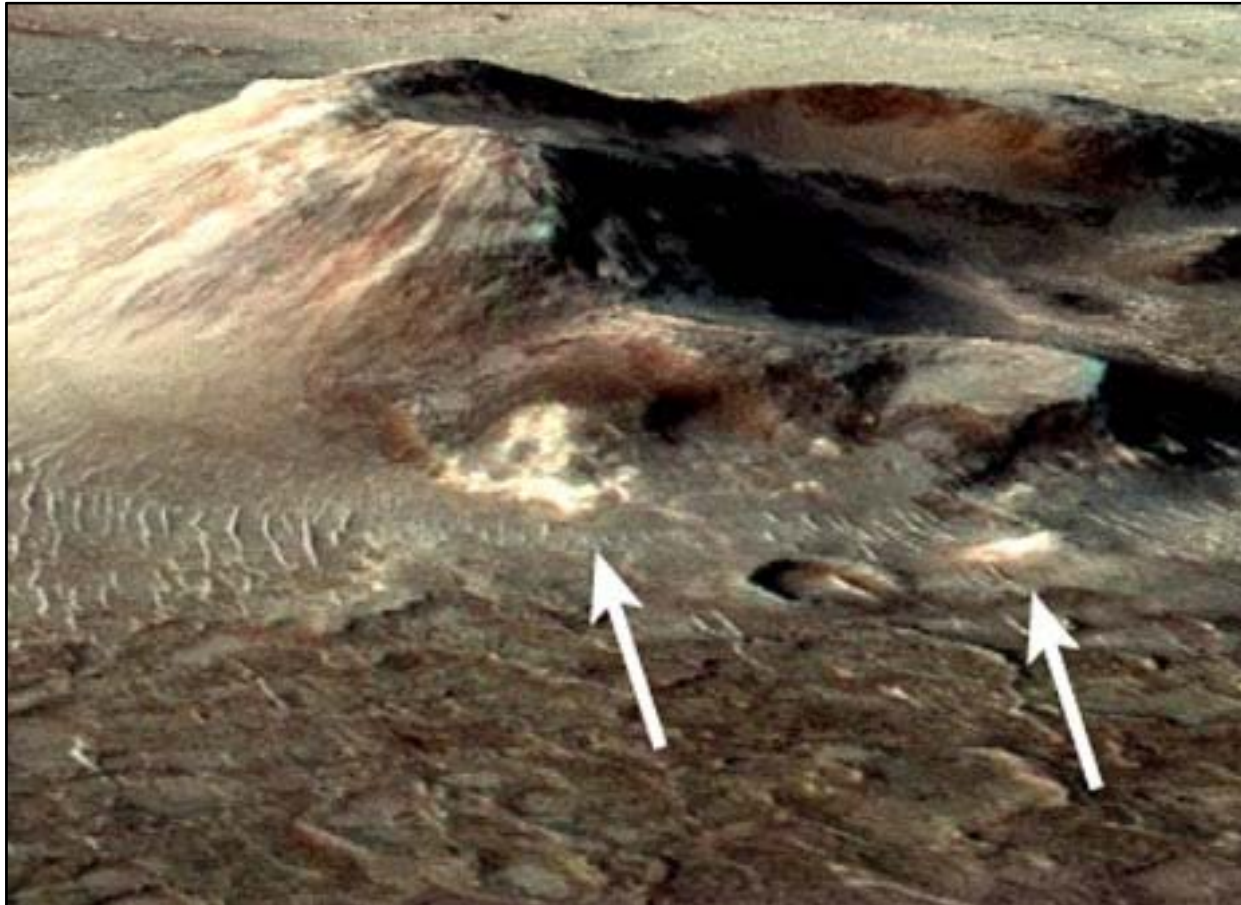


Volcanism



Impact





- Nili Patera caldera in Syrtis Major hosts deposits of hydrated silica that are spatially related to volcanic cones (Skok *et al.*, 2010).

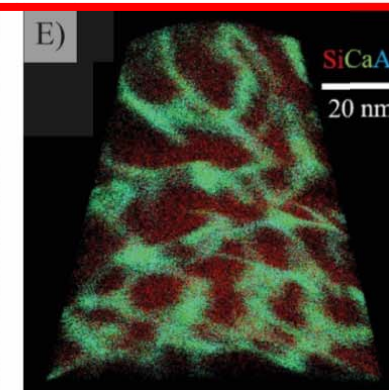
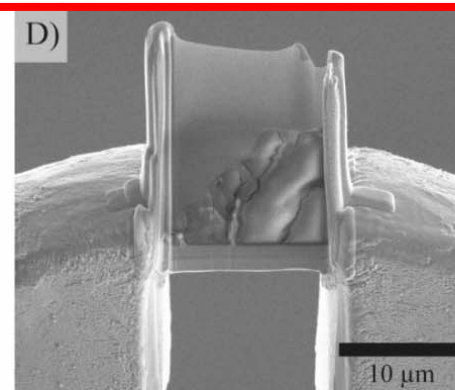
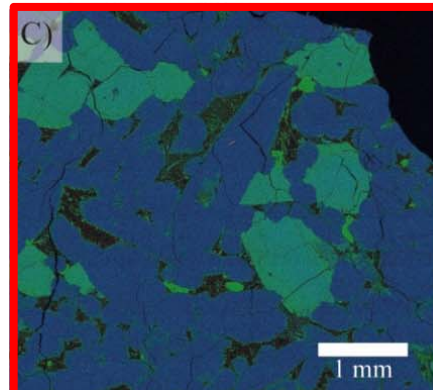
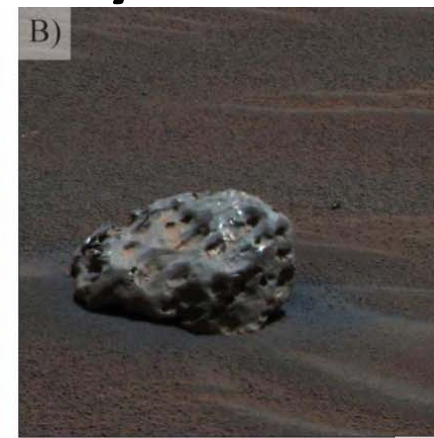
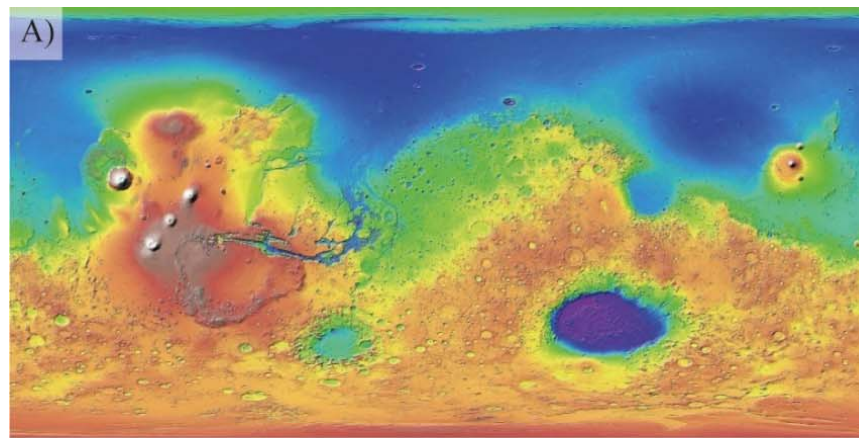
Nakhlite meteorites history: Impact-ejection

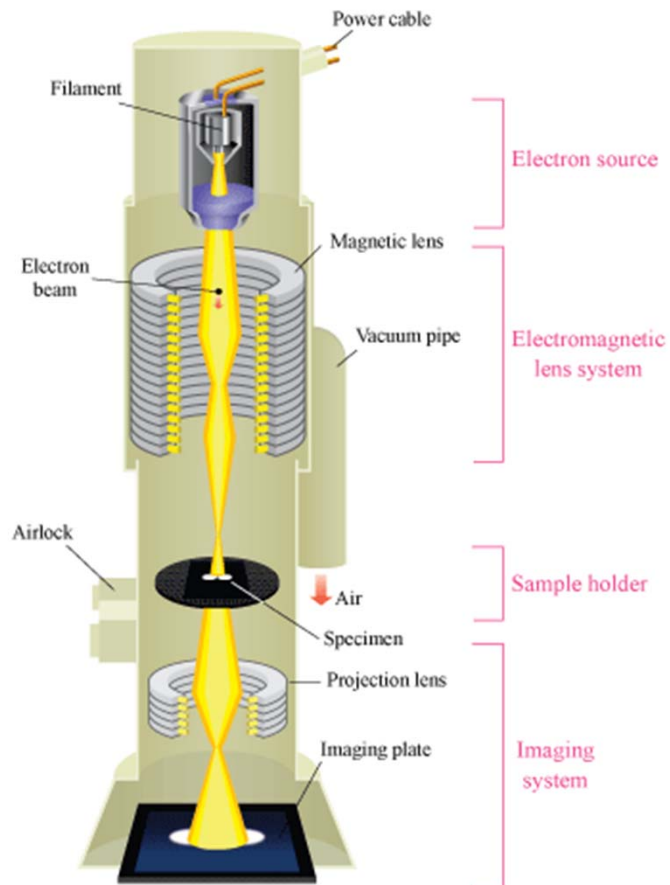


~1-20 million years



Correlative Study

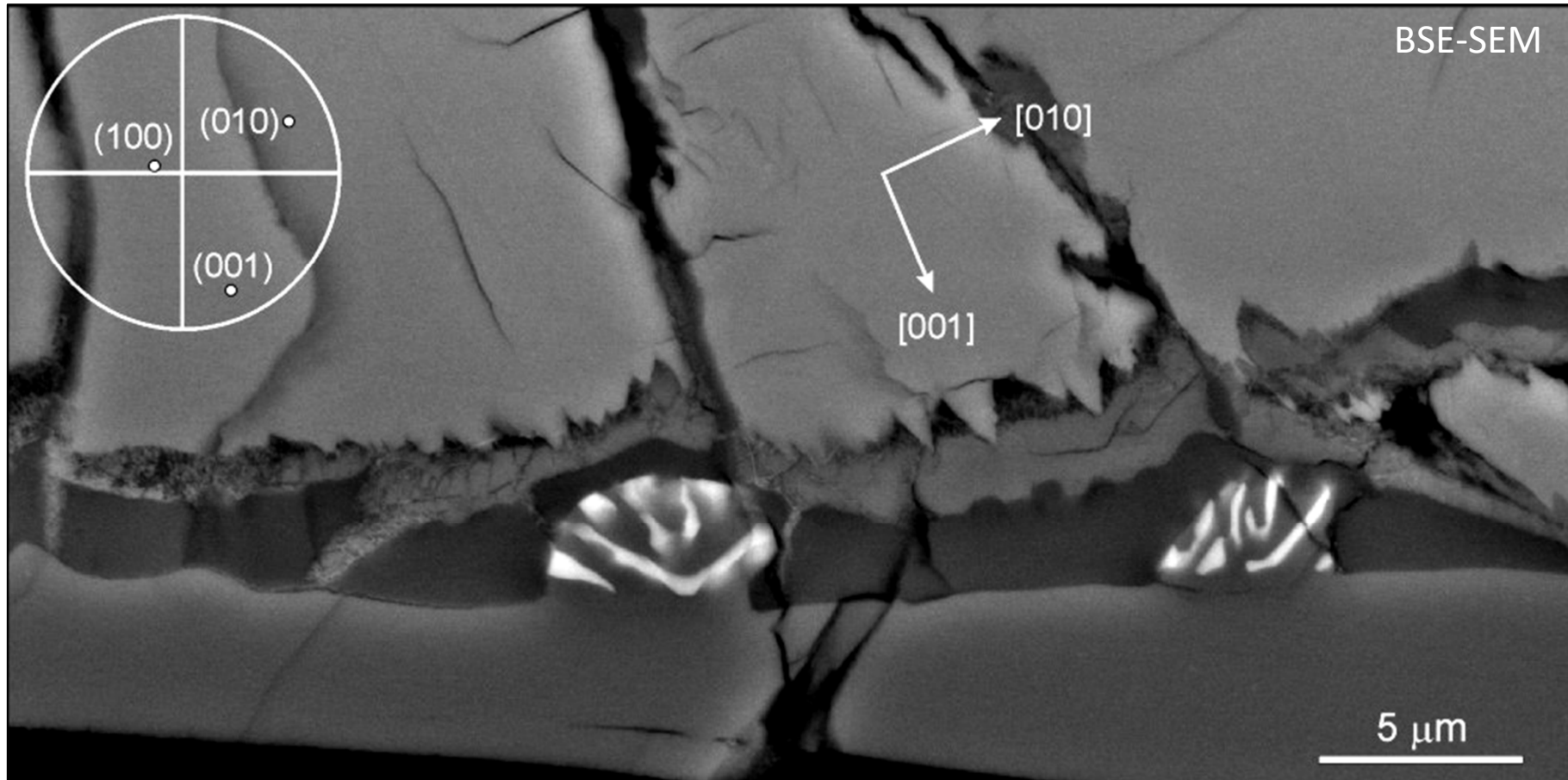




- JEOL ARM200F. Nominal spot size in STEM mode 0.2 nm.
- EELS using a Gatan GIF Quantum ER spectrometer;
- EDX using a Bruker SDD/EDX spectrometer.



SEM: Water-rich 'iddingsite' veins within olivine



2015



Amorphous gel

Changela & Bridges

2005

Smectite, hematite, goethite, quartz,
cristobalite, tridymite, amorphous silicaThomas-Keptra *et al.*

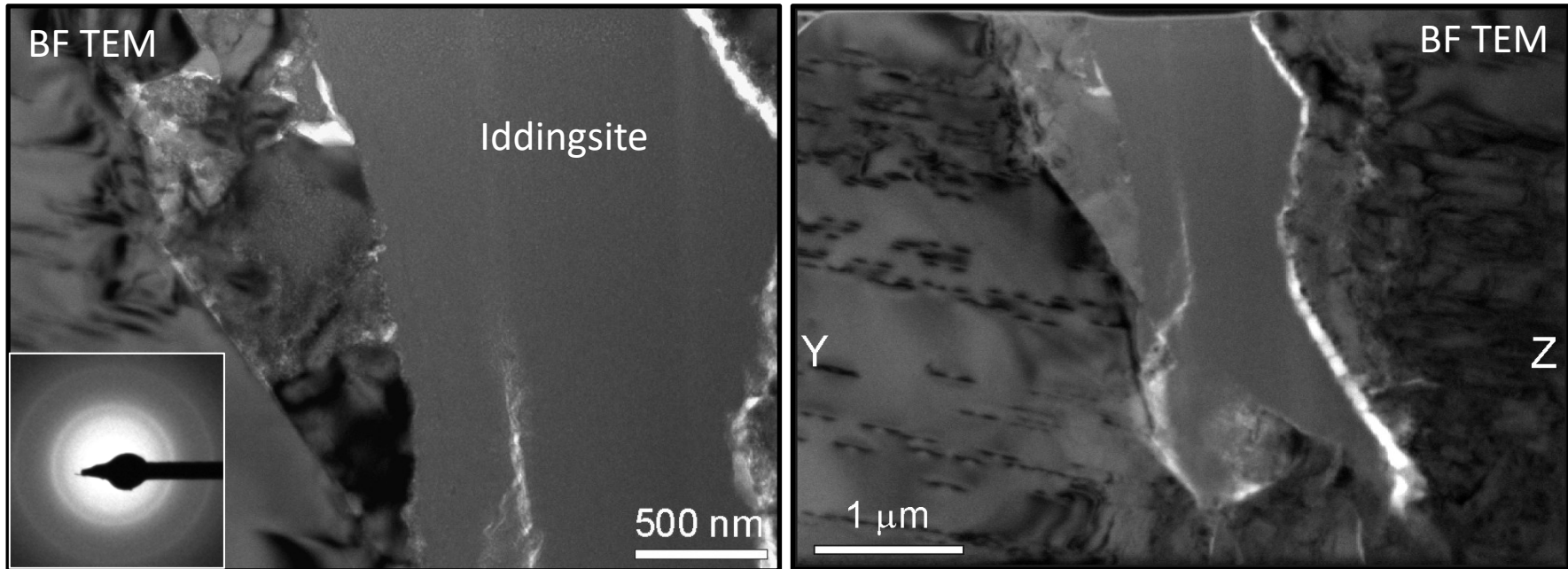
1995

Fe-saponite, hematite, two-ring ferrihydrite
SmectiteTreiman & Gooding
Gooding *et al.*

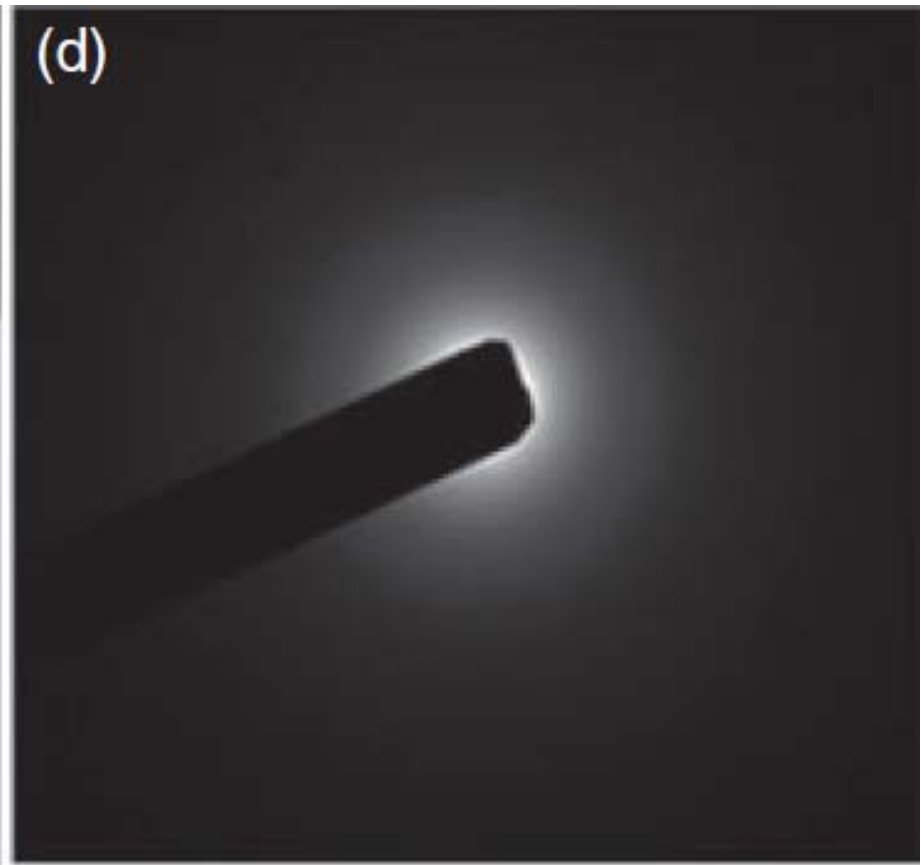
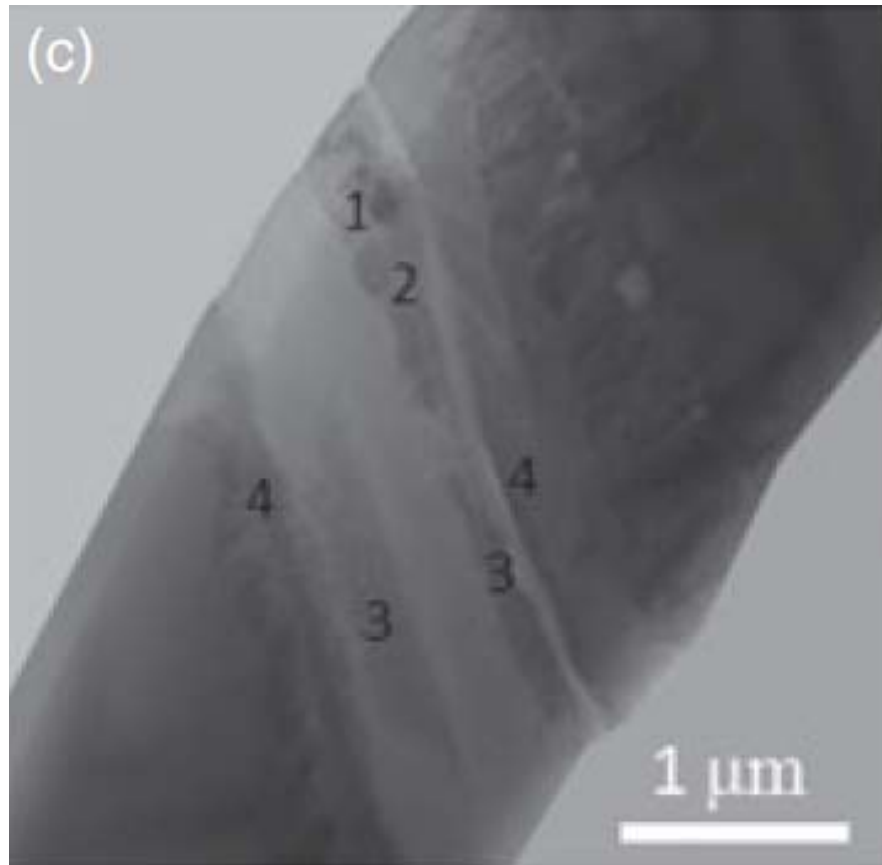
1985

1975 Fine-grained/colloidal material

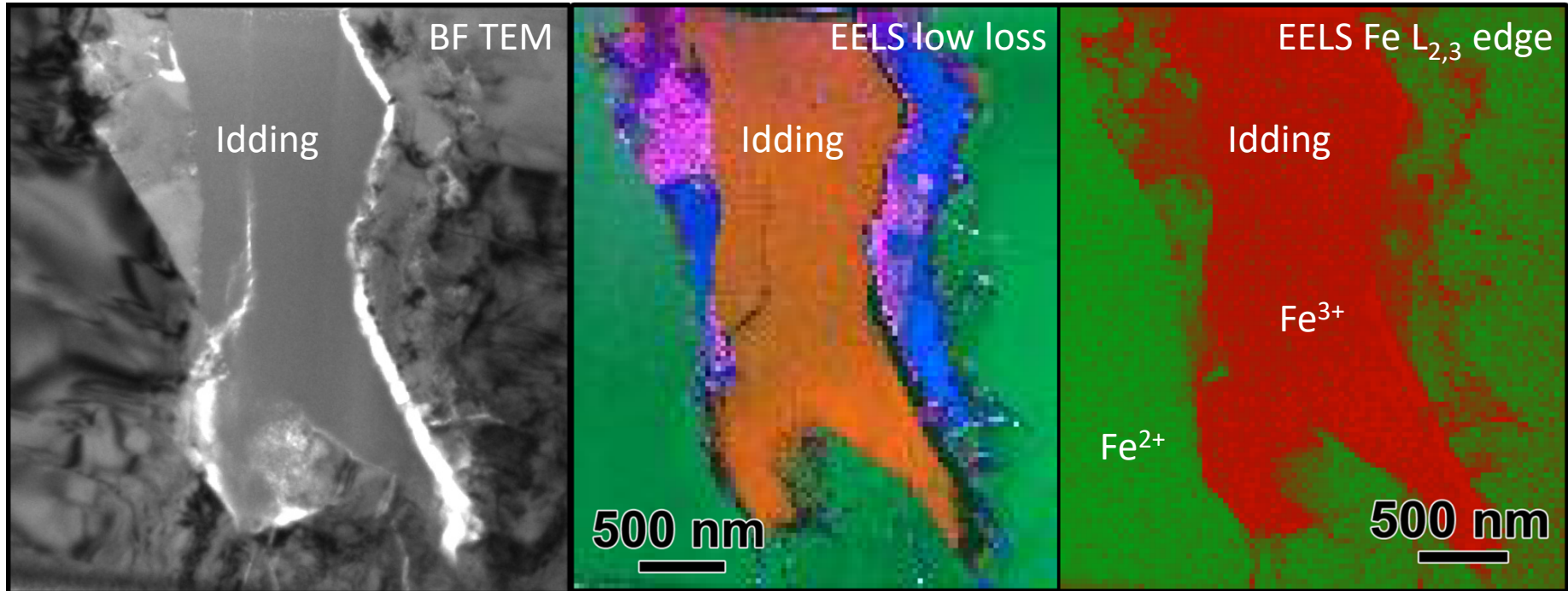
Ashworth & Hutchison



- Foils made by FIB, then Ar ion milled to ~50 nm.
- Iddingsite is featureless in BF TEM and nanocrystalline.

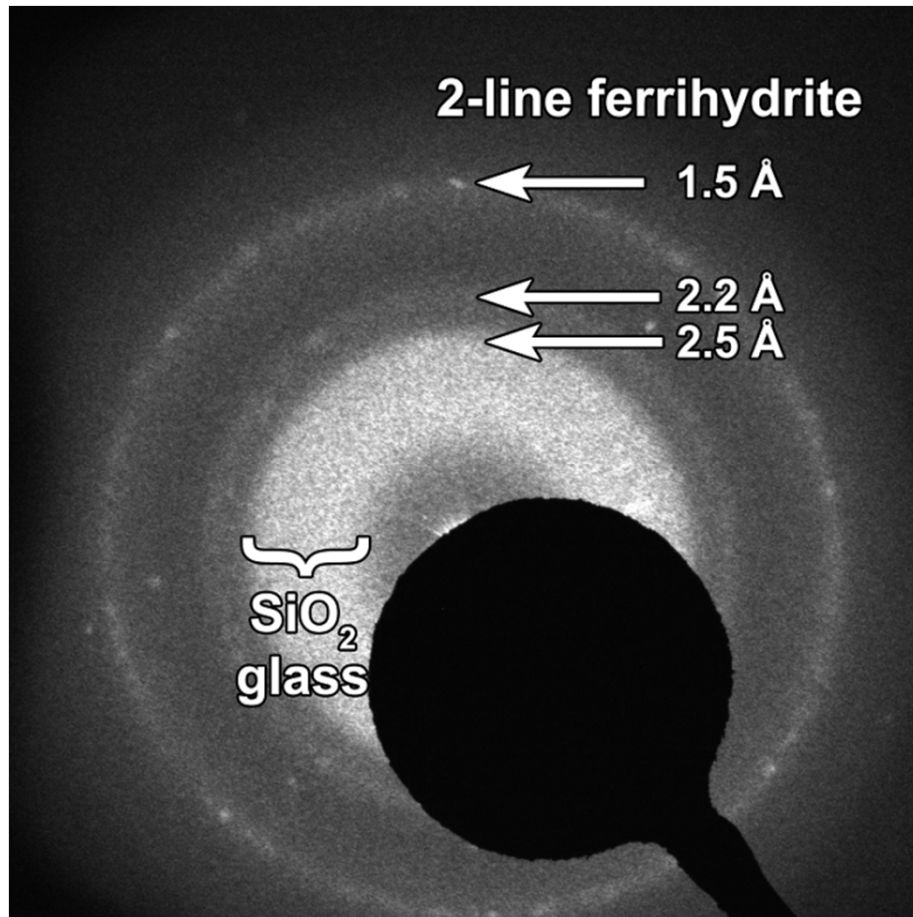


TEM : Vein phase identification using EELS

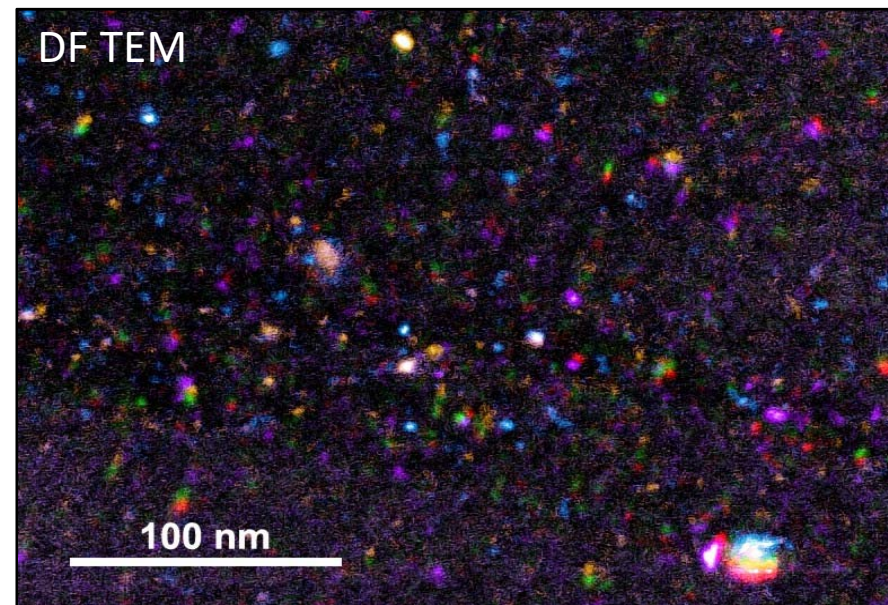


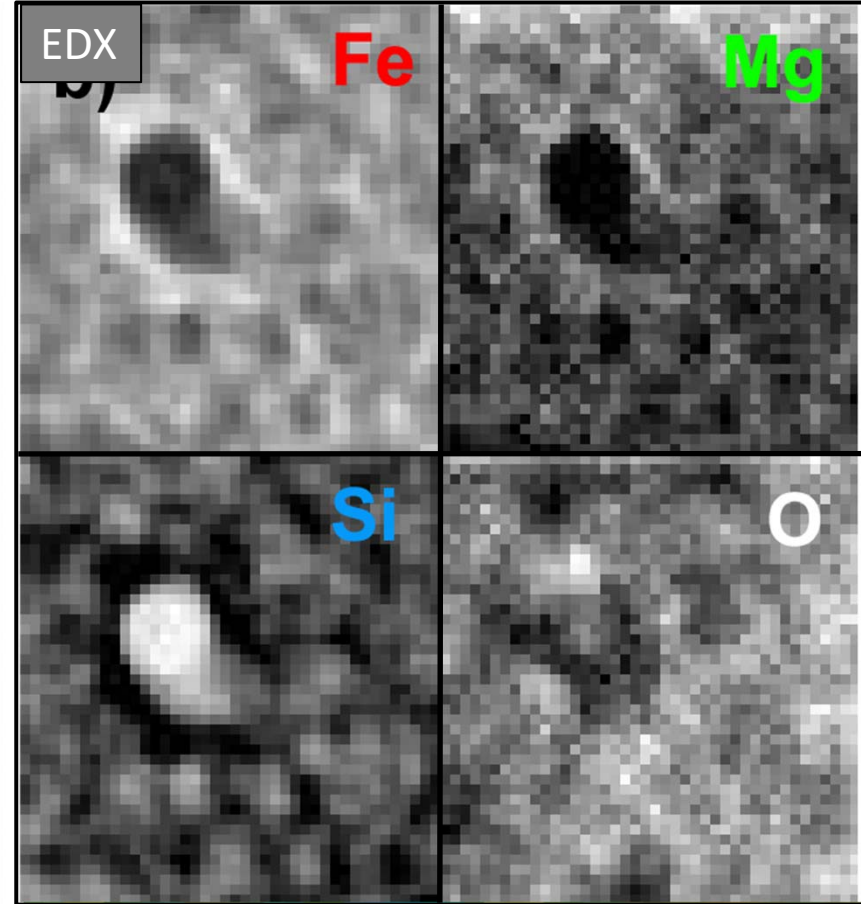
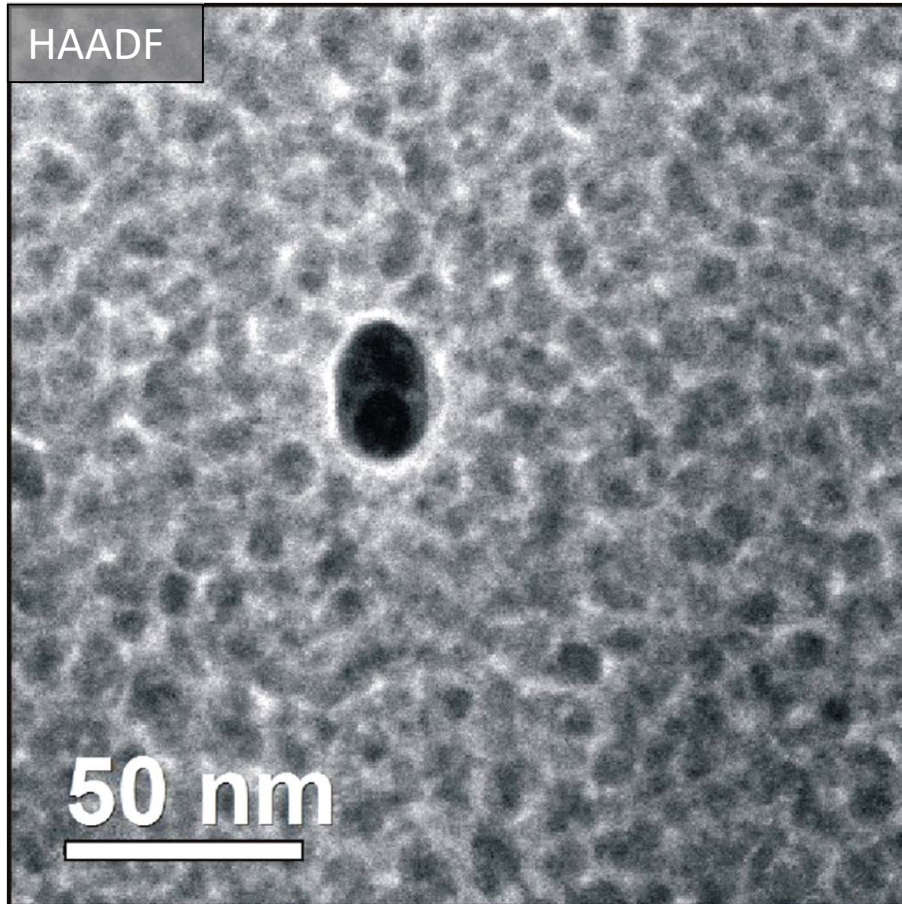
Olivine: $(\text{Fe}^{2+}, \text{Mg})_2\text{SiO}_4$
Siderite: $\text{Fe}^{2+}\text{CO}_3$
Goethite: $\text{Fe}^{3+}\text{O}(\text{OH})$

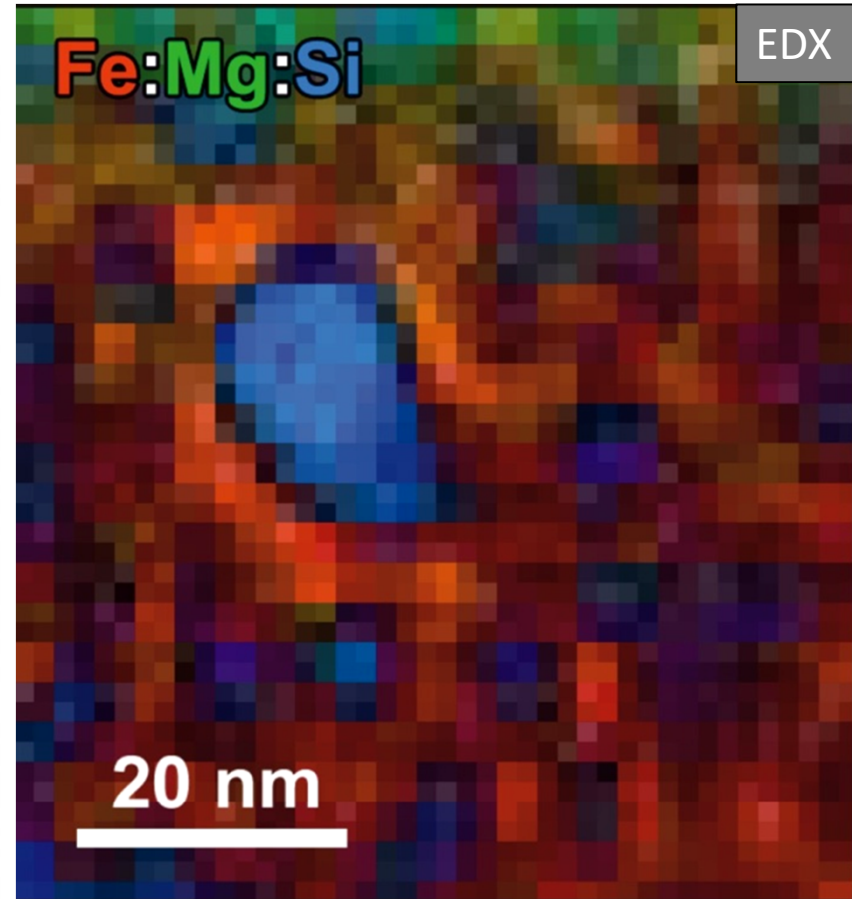
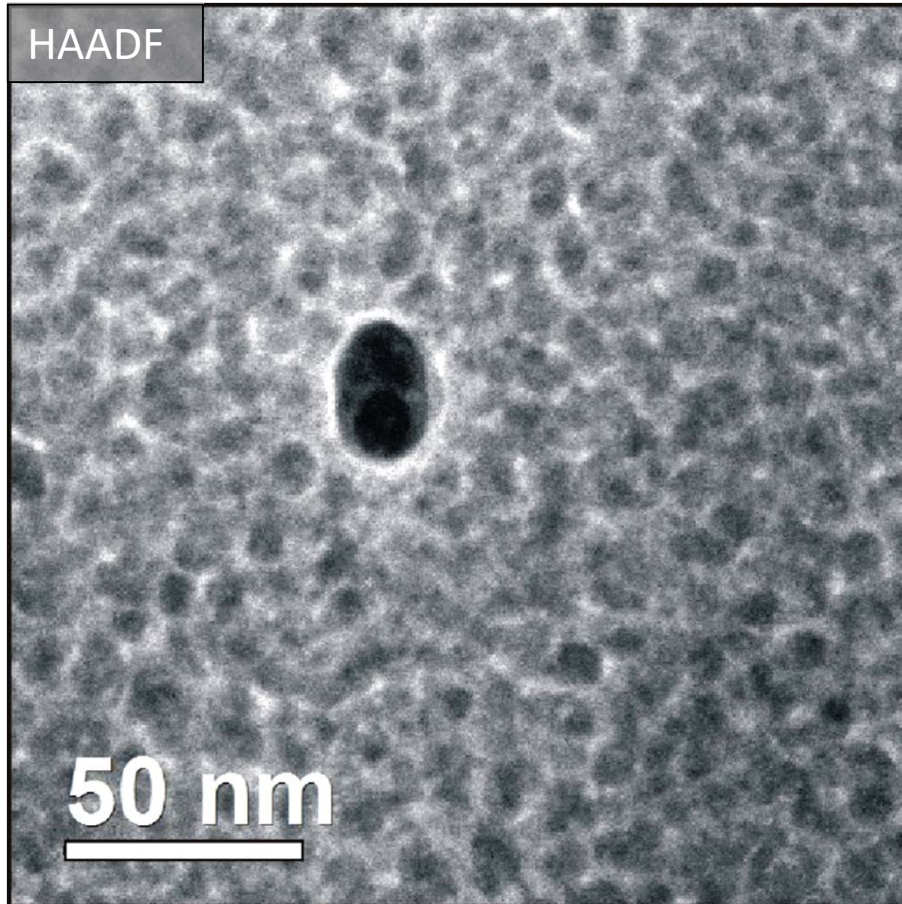
TEM : Water-bearing constituents of iddingsite



- SAED and dark-field TEM reveals nanocrystals of ferrihydrite:
 $\text{Fe}^{3+}_{8.2}\text{O}_{8.5}(\text{OH})_{7.4} + 3\text{H}_2\text{O}$.







TEM : The carriers of water identified

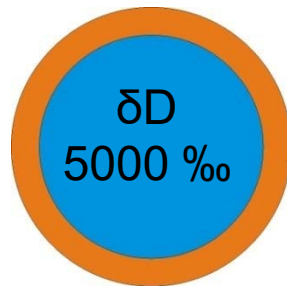
- Nanoparticles of opal-A $\text{Amorphous SiO}_2 \cdot n\text{H}_2\text{O}$
- Shells of ferrihydrite $\text{Fe}^{3+}_{8.2}\text{O}_{8.5}(\text{OH})_{7.4} + 3\text{H}_2\text{O}$
- Shells also host nontronite? $(\text{Ca},\text{Mg},\text{Na})_{0.3}\text{Fe}^{3+}_2(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot n\text{H}_2\text{O}$

Terrestrial 'fire opal'



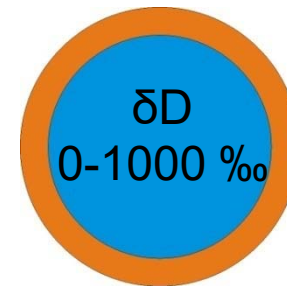
- δD of opal is lower than the present-day Martian atmosphere due to exchange of H whilst on Earth;
 - The nanoparticles have a reactive surface area of 250 m²/g.
Industrial catalysts = 1-1000 m²/g.

Mars atmosphere
 δD 5000 ‰

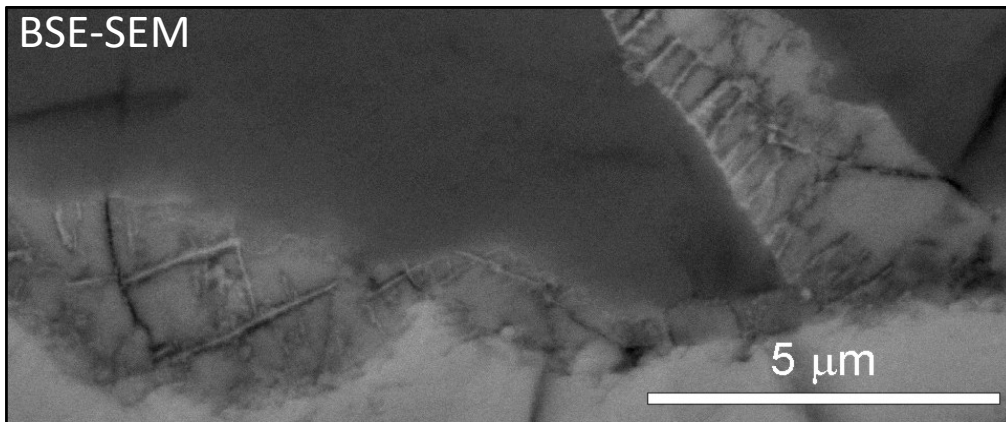
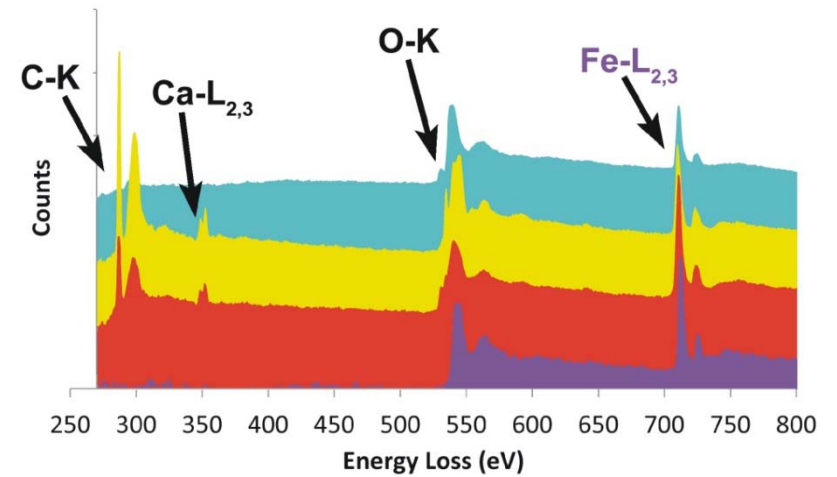
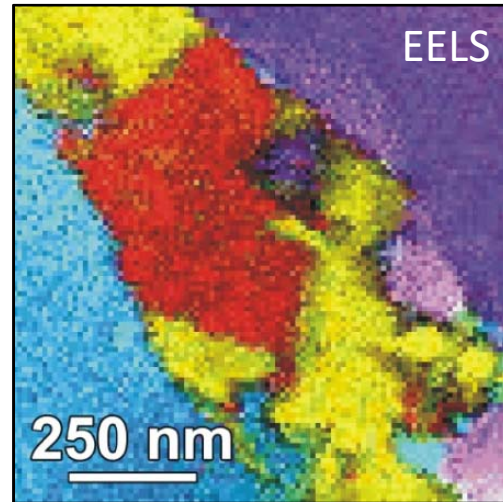
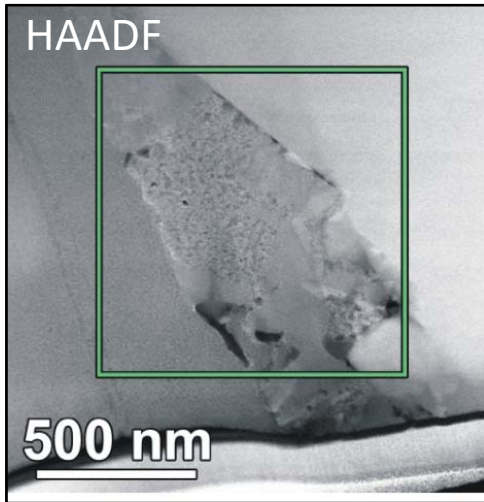


 Opal-A
 Ferrihydrite

Earth atmosphere
 δD 0 ‰

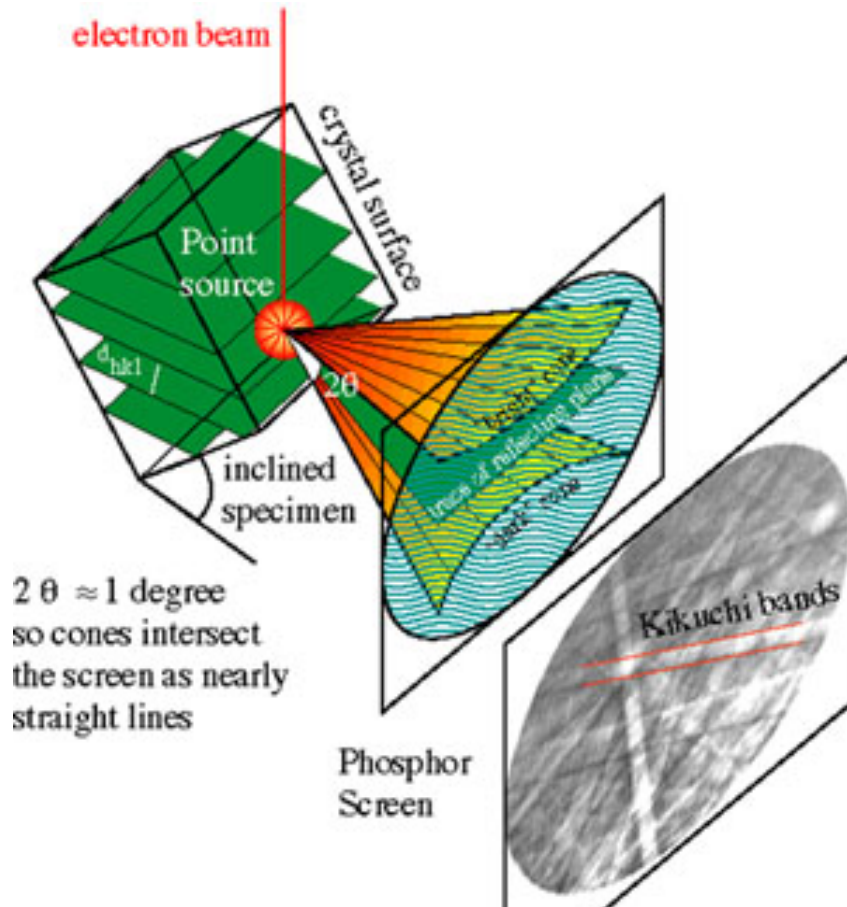


Addition & exchange of water/OH whilst on Earth



Siderite Goethite Olivine Iddingsite

- Goethite formed by oxidation of siderite. Maybe after fall of the meteorite.



<http://www.gm.univ-montp2.fr/spip.php?article104&lang=fr>

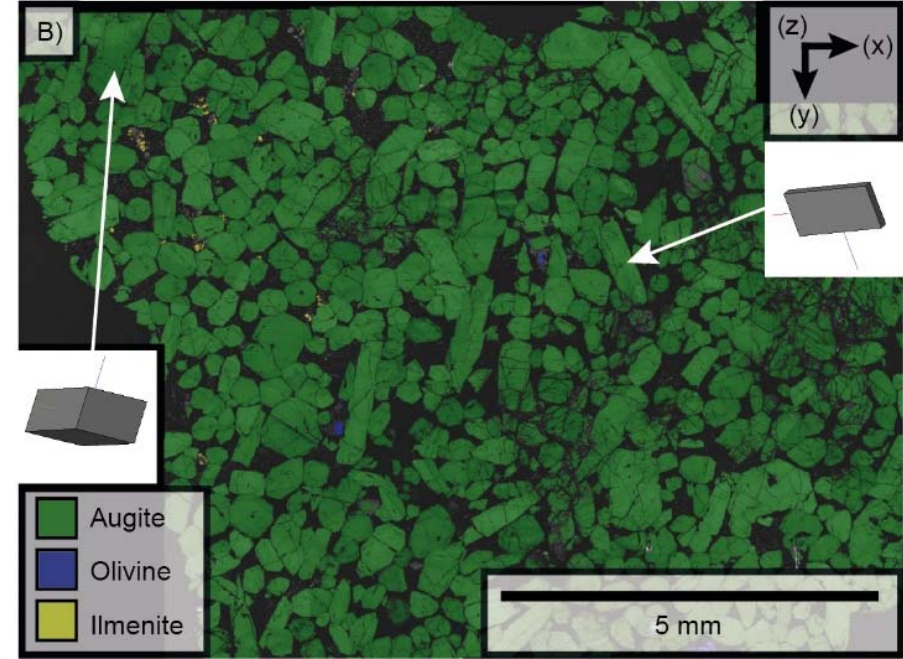
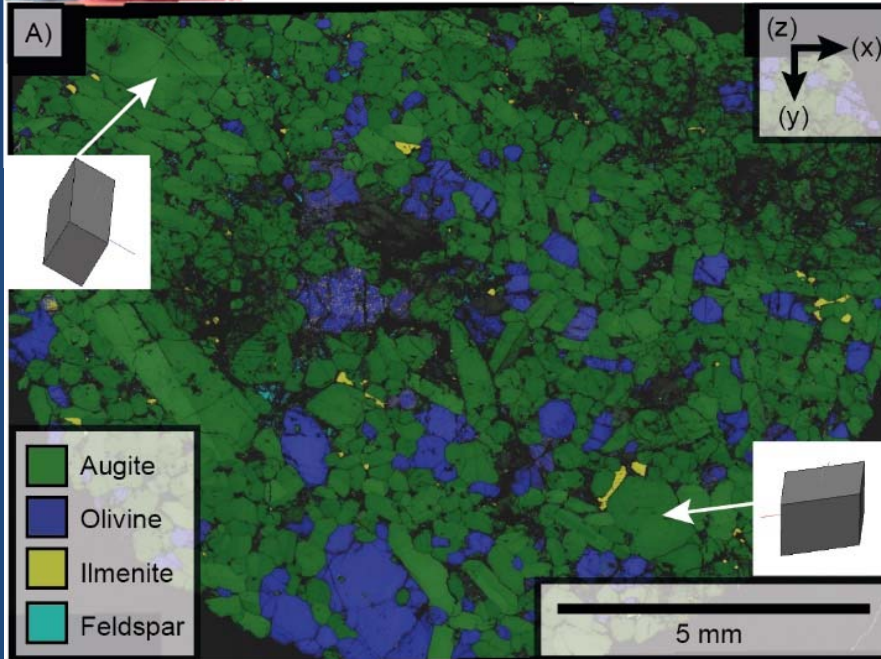
- Rapid crystallographic mapping
- Max spatial resolution ~80 nm
- Max angular resolution ~0.1°
- Mineralogy
- Crystallinity
- Crystal orientation
- Internal deformation
- Grain shape
- Zeiss Sigma VP – Oxford Instruments



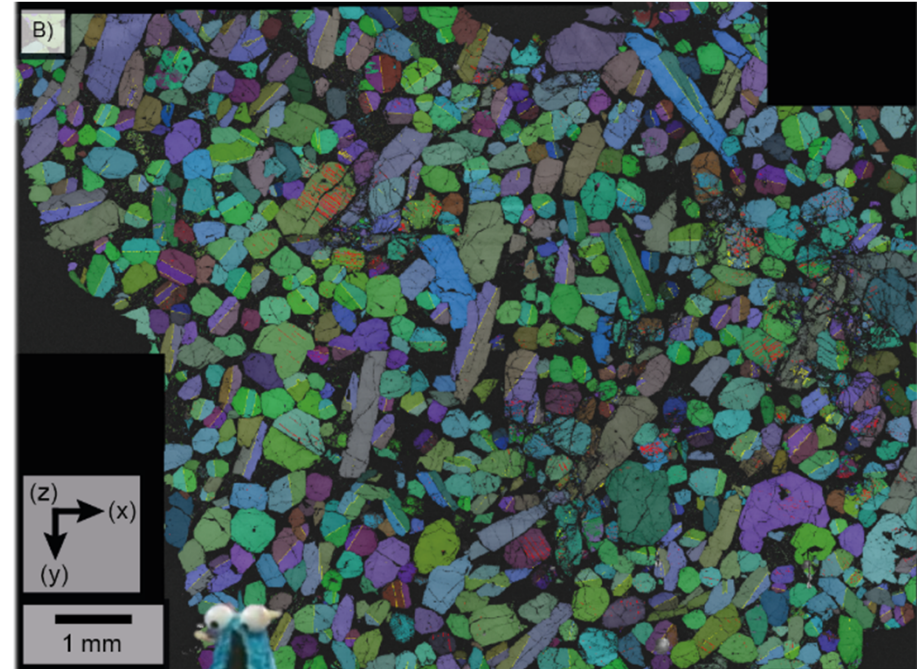
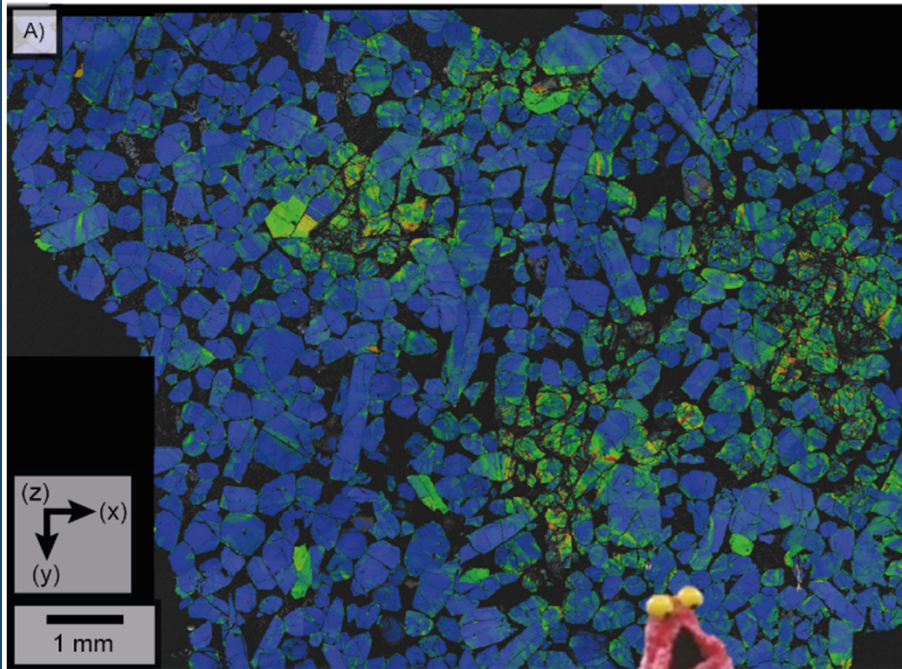


Lafayette

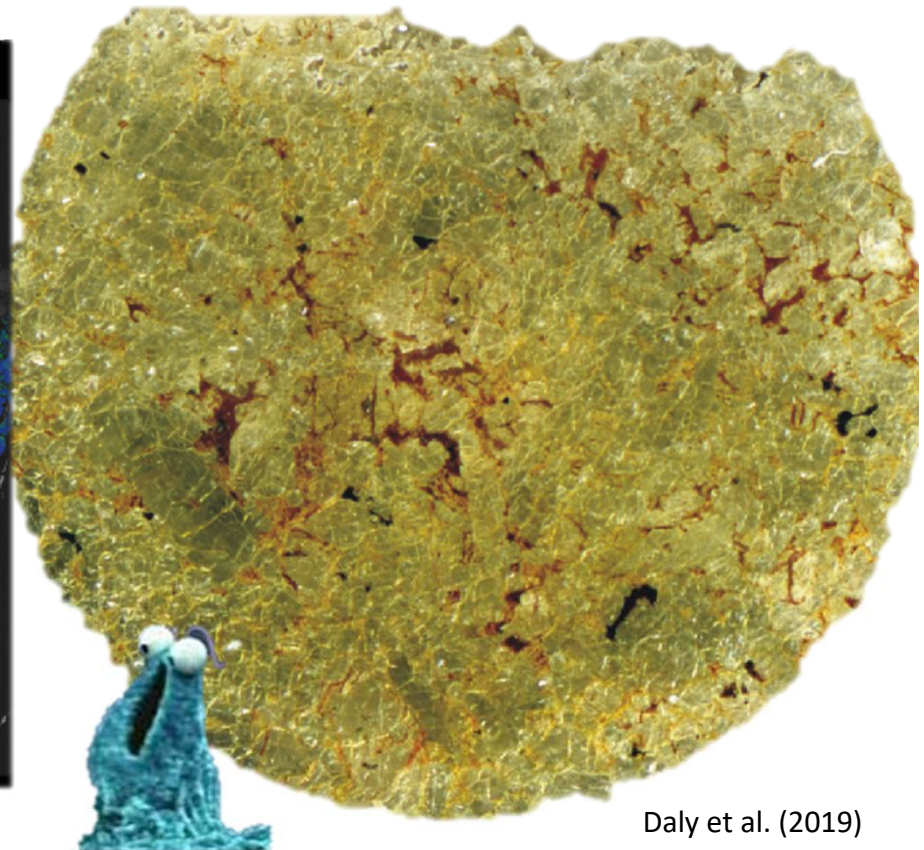
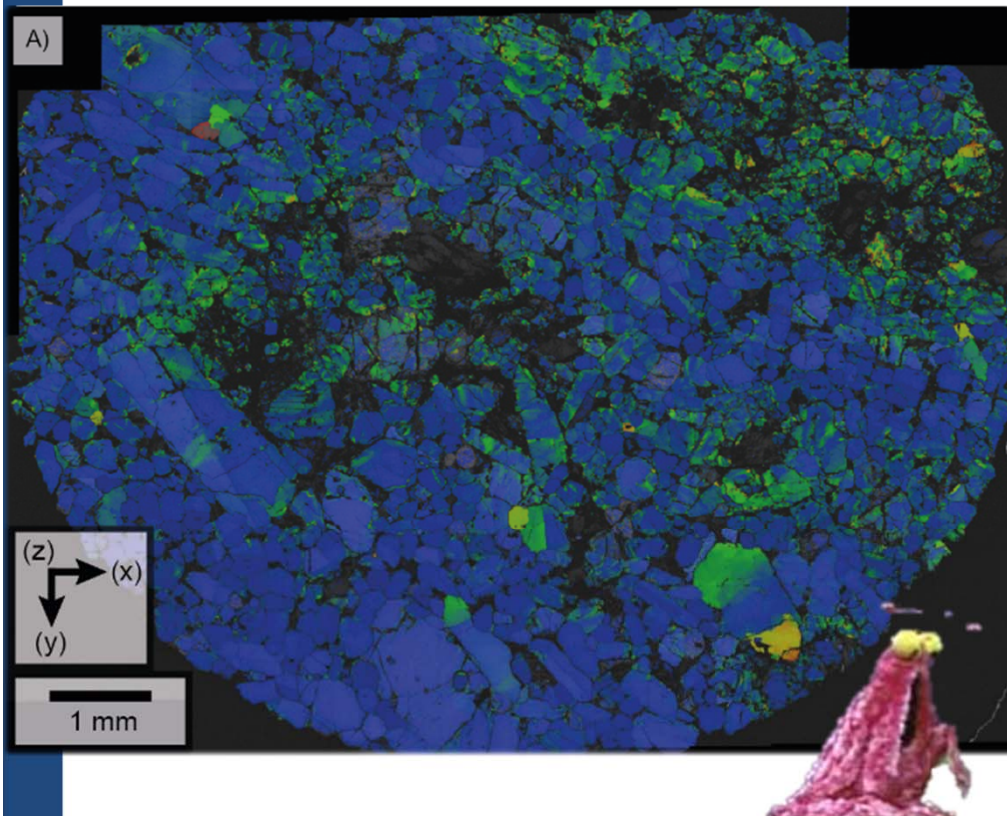
MIL 03346

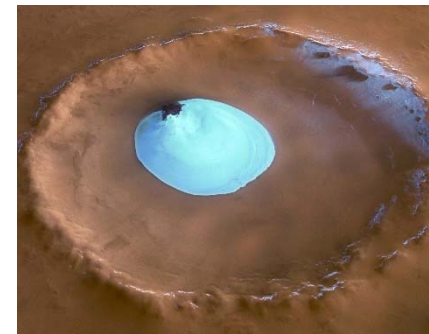
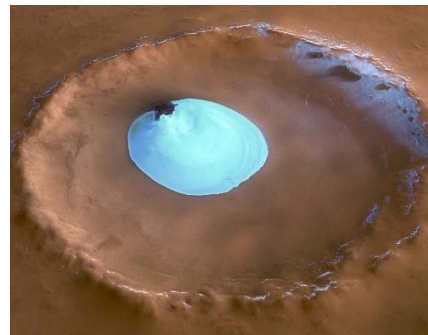


MIL 03346



Lafayette





1.4-1.3

0.633

0.011

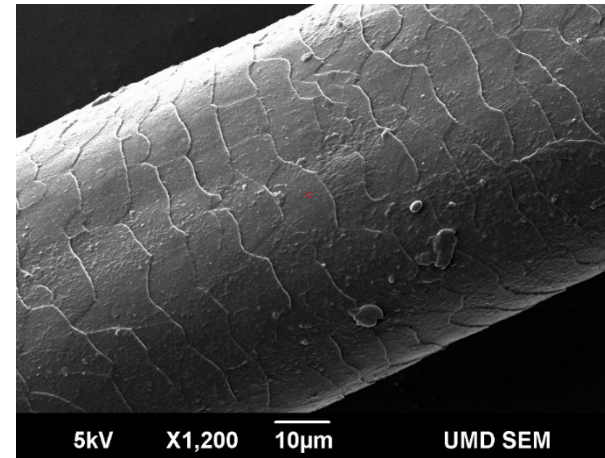
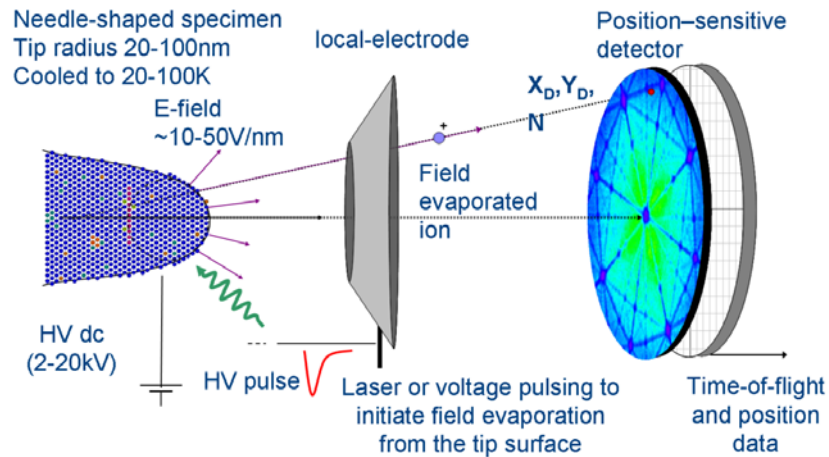
0

Billion Years

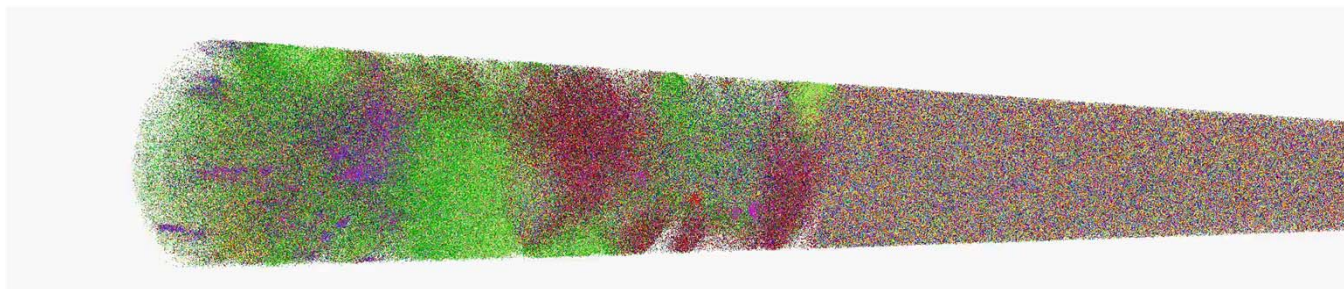
APT : searching for robust water reservoirs



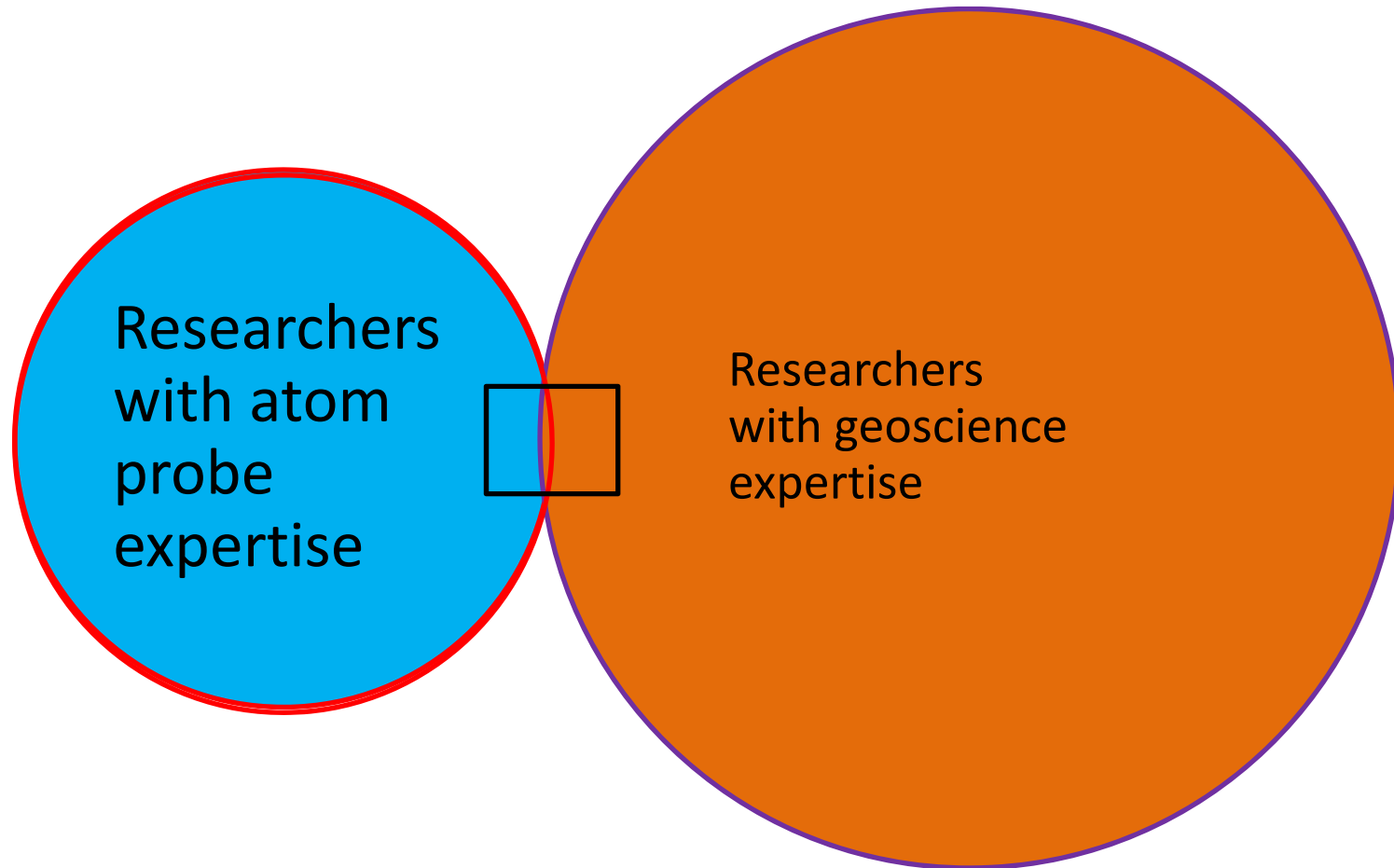
- LEAP 5000XHR Atom Probe, University of Oxford.
- Detection limits ~ 1 atomic ppm
- 100% ionisation - Whole periodic table
- Molecular species including OH and H₂O
- Mass spectrum and 3D atomic positions reconstructed using the IVAS software.



Gault et al. 2012



3D, Geochemistry, Isotopes





University
of Glasgow

Atom Probe in Planetary Science

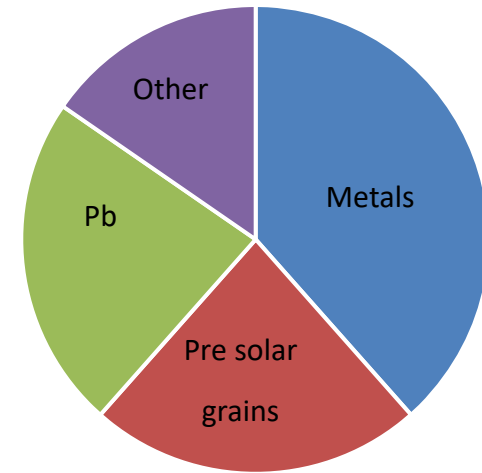
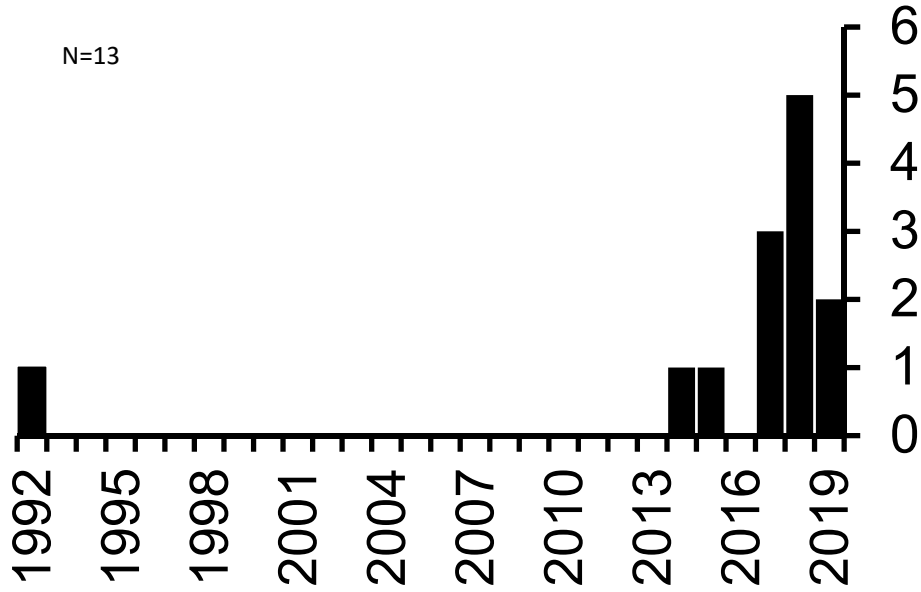
Researchers
with atom
probe expertise

Researchers with
planetary
science
expertise

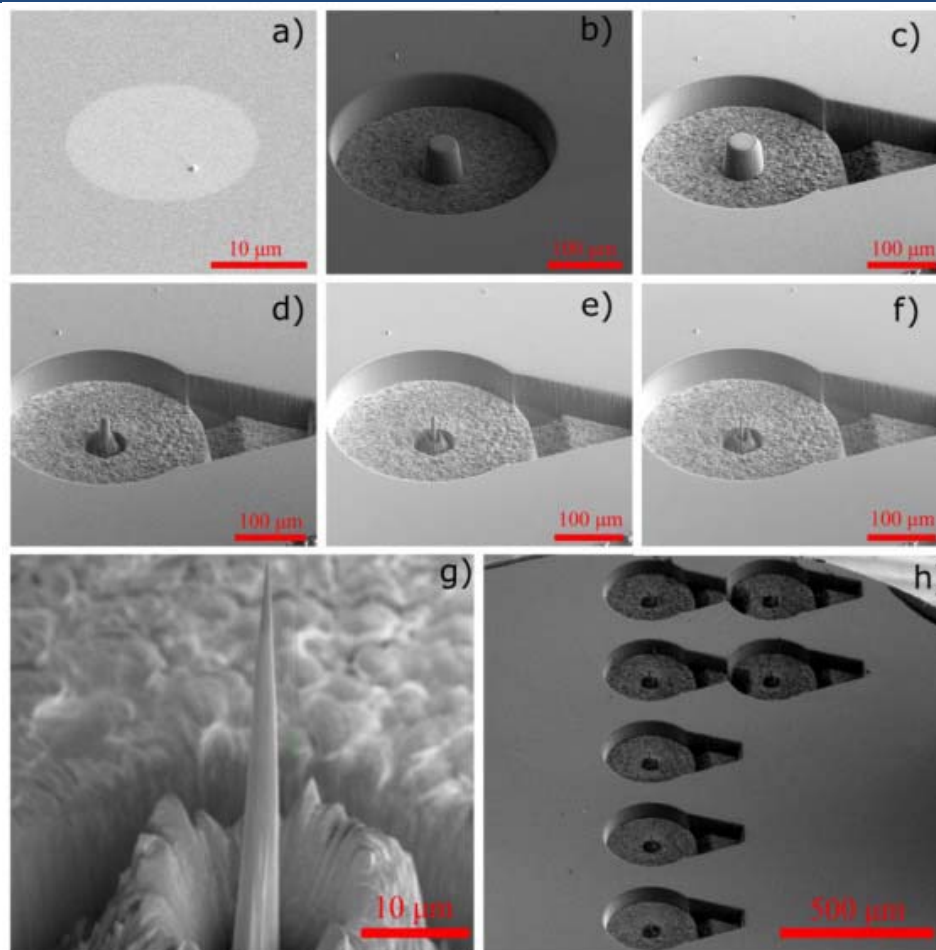
Researchers
with geoscience
expertise



Atom probe Planetary Science publications



Site specific: Focused ion beam sample preparation

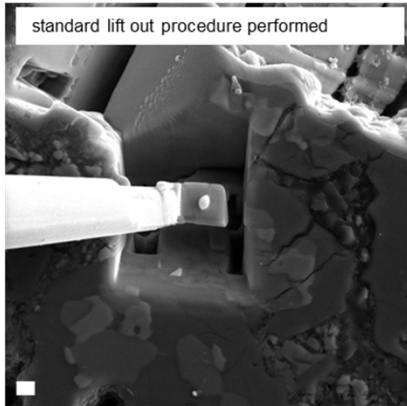
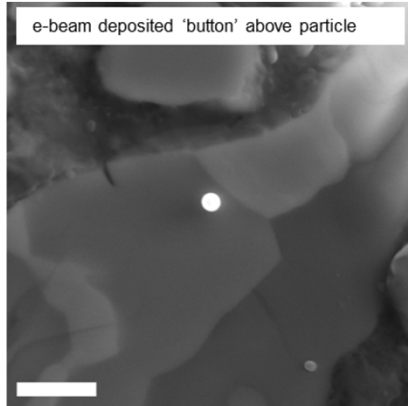
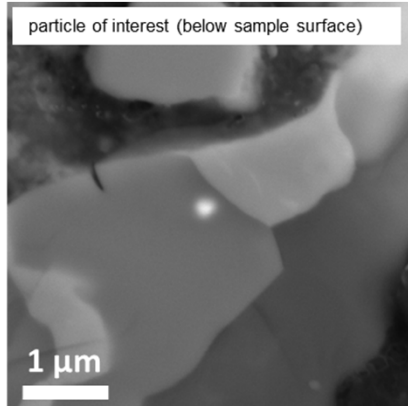


Satellite dish
method



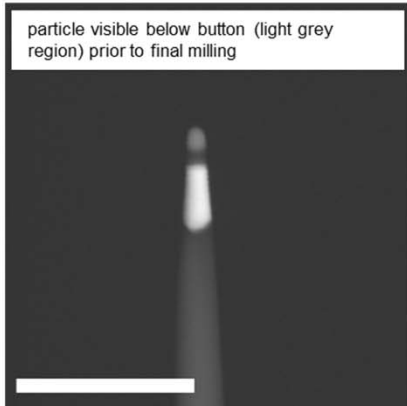
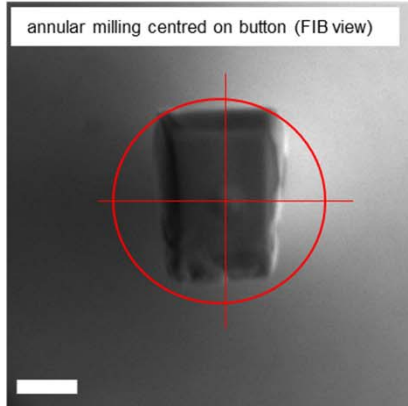
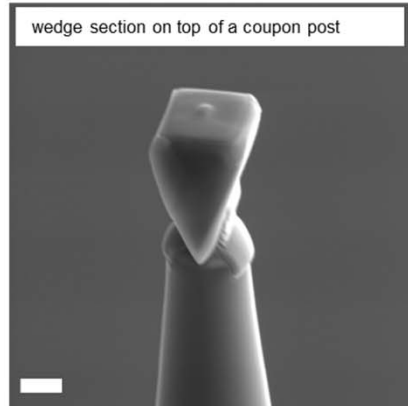
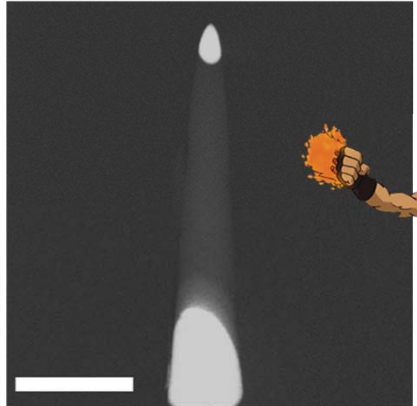
Halpin et al., 2019

Site specific: Focused ion beam sample preparation



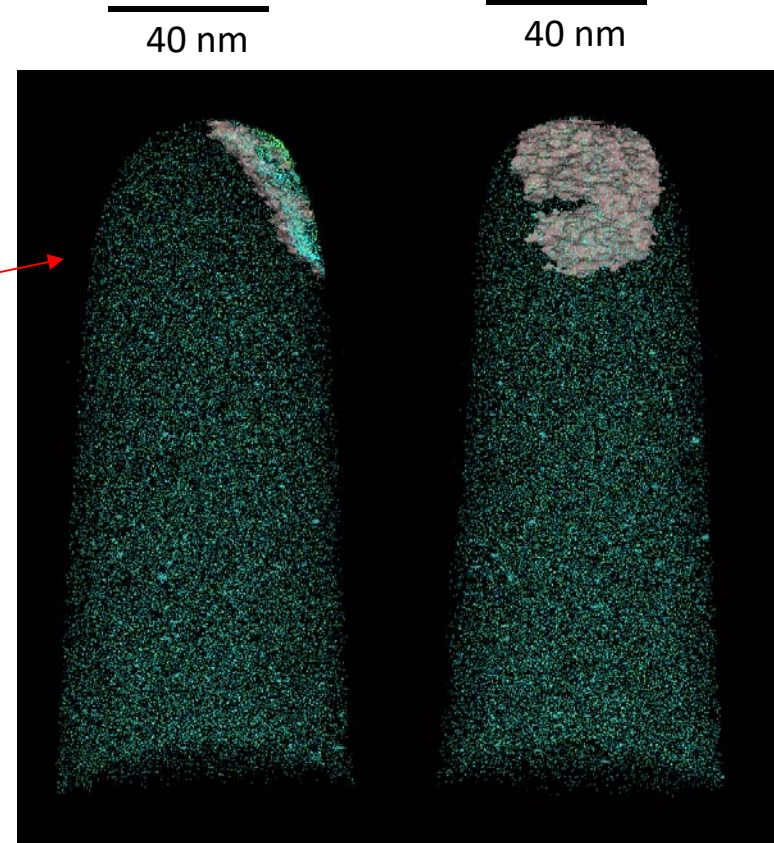
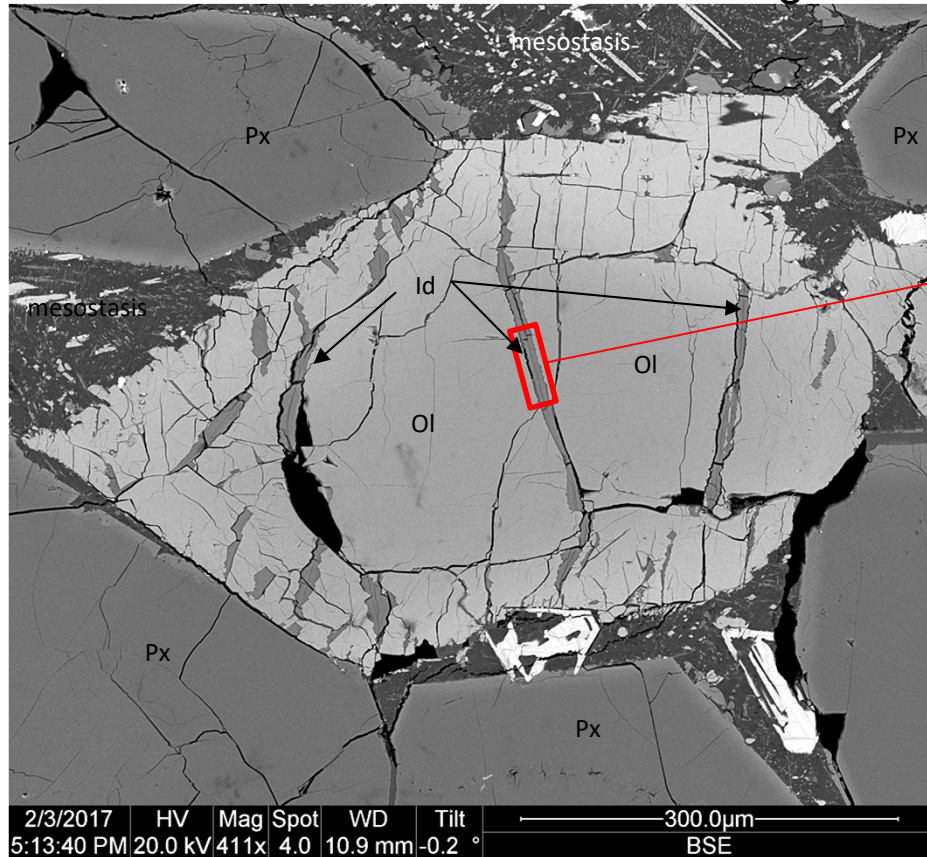
Button method

Particle successfully located at tip of APT needle

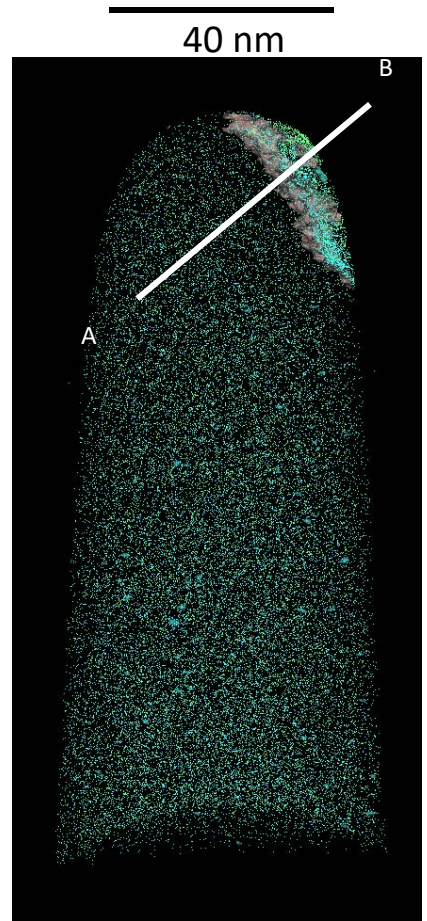


Rickard et al., (in review)

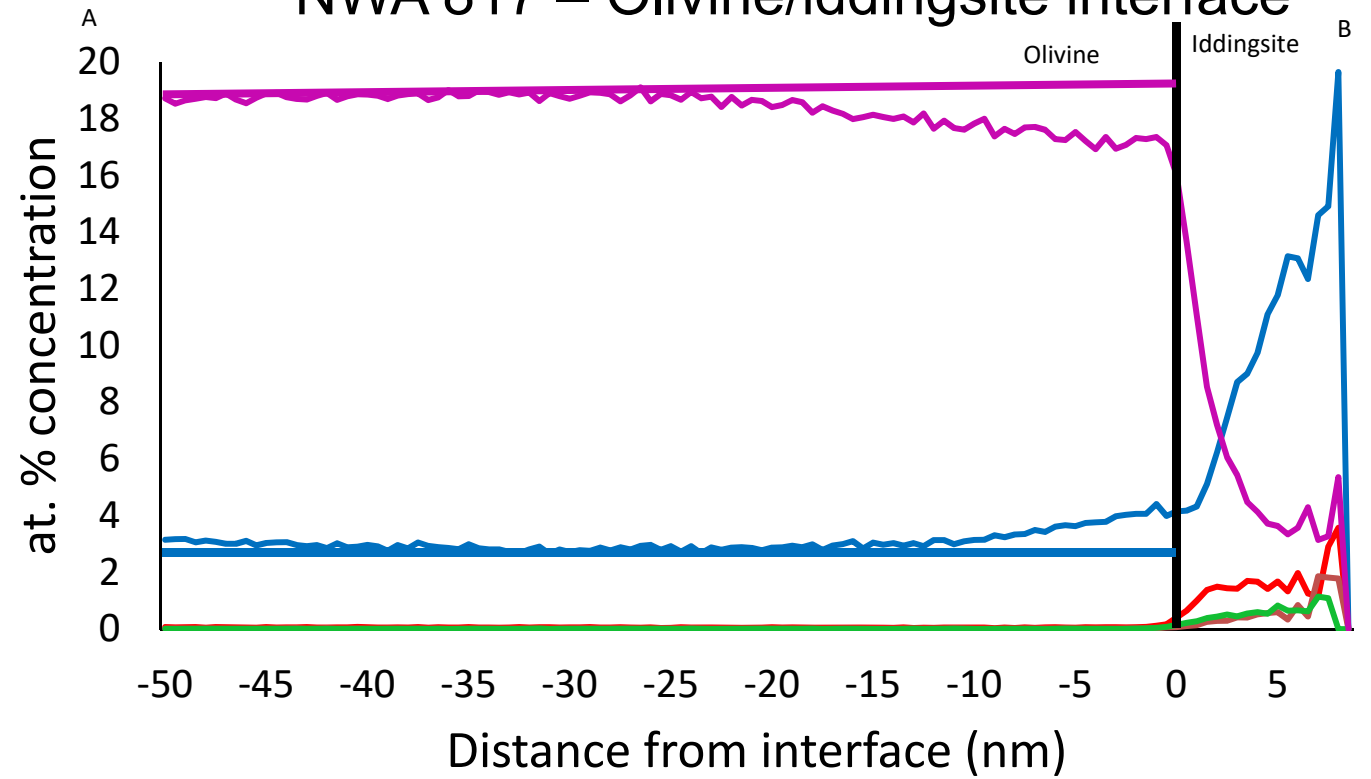
NWA 817 – Olivine/iddingsite interface



APT : searching for robust water reservoirs



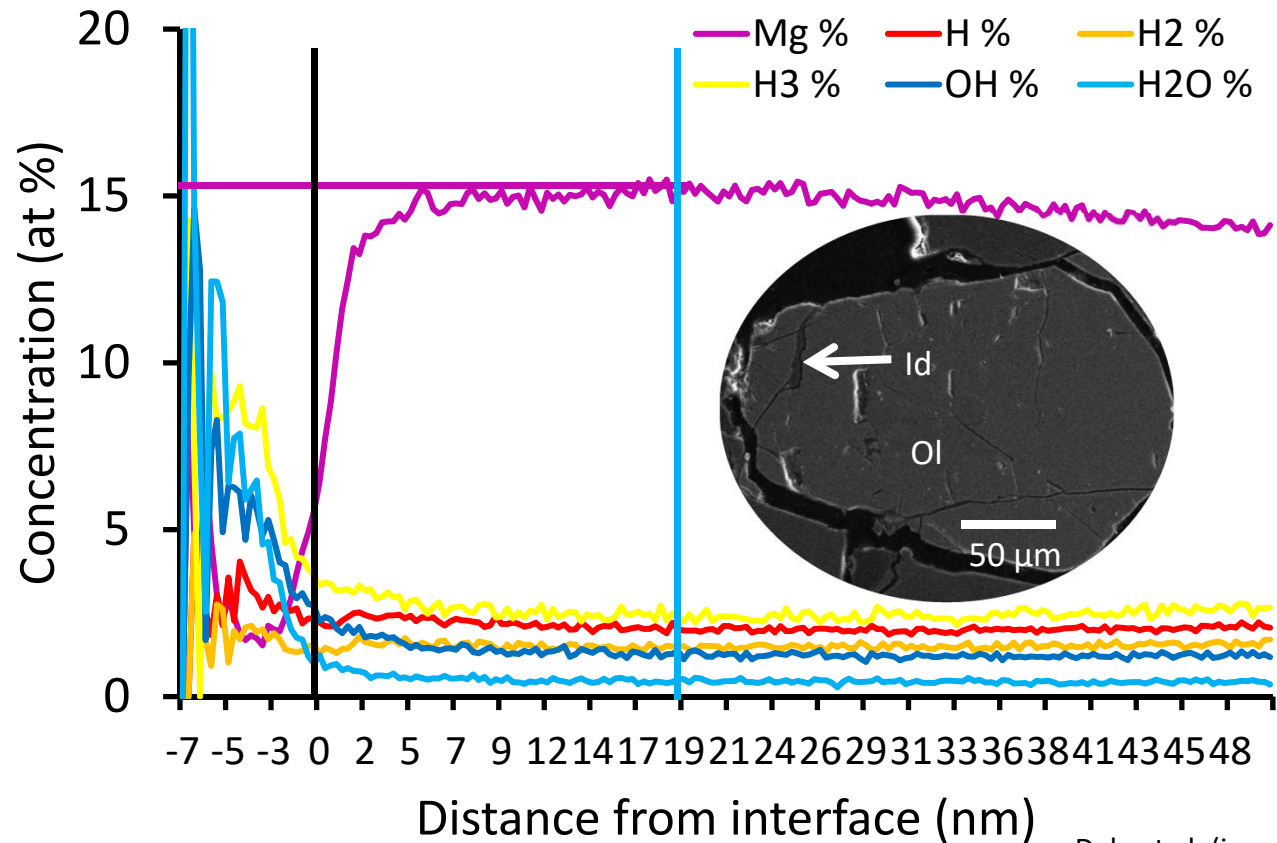
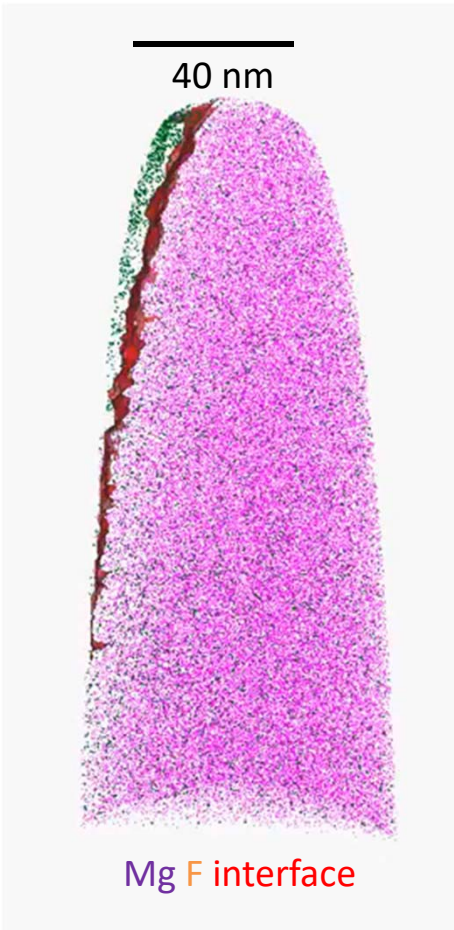
NWA 817 – Olivine/iddingsite interface



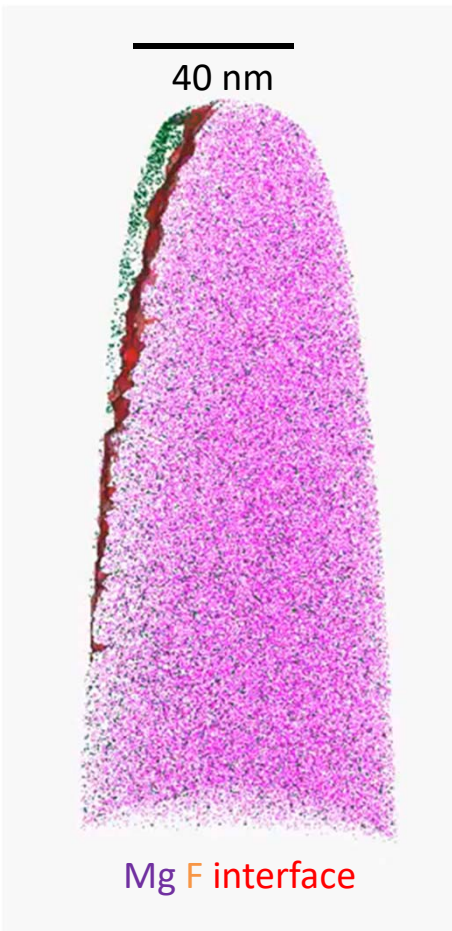
— Al — Na — H — As — Mg

APT : searching for robust water reservoirs

Lafayette Olivine/iddingsite interface

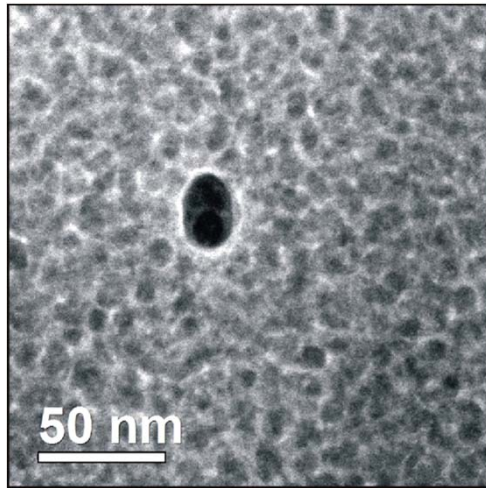


APT : implications for water reservoirs



- Water is locked into the reaction rim in the olivine at the olivine-iddingsite interface.
- Water rock reaction occurs through cation exchange not dissolution-reprecipitation
- δD of this water exchanges less readily with the terrestrial atmosphere
 - Could provide a more robust Martian δD measurement to determine the provenance of the water
 - Meteoric vs crustal vs magmatic vs impact





- A impact event preceded and may have driven fluids on Mars ~633 Ma
- Water is hosted in opal-A ('fire opal'), and olivine-iddingsite reaction rims.
- The nanoparticles, and possibly also siderite have been altered by the Earth's atmosphere.
 - But olivine rims may be resistant.
- Returning samples for analysis by techniques including EBSD/TEM/APT is essential.





University of Glasgow

Thanks to my collaborators

Imperial College London



Curtin University



THE UNIVERSITY OF SYDNEY



UNIVERSITY OF LEEDS



MONASH University



MACQUARIE University



University of Strathclyde



UNIVERSITY of HAWAI'I MĀNOA



UNIVERSITY OF CAMBRIDGE



LUNDS UNIVERSITET



Smithsonian Institution



THE UNIVERSITY OF WESTERN AUSTRALIA





University
of Glasgow

Annual Meeting
of the Meteoritical
Society

9th-14th Aug 2020

